

# দিনাজপুর বোর্ড

- ১৩। কোনটির ক্ষিপ্ততা সবচেয়ে বেশি?  
(ক)  $\text{H}_2\text{SeO}_3$  (খ)  $\text{HBrO}_4$   
●  $\text{HClO}_4$  (গ)  $\text{HIO}_4$

- ১৪। স্রোতার দ্রবণের সংকেত কোনটি?  
(ক)  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$   
(খ)  $\text{FeSO}_4 \cdot 7\text{H}_2\text{O}$   
(গ)  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$   
●  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$

- ১৫। কোনের দ্রবণের  $\text{pH}$  12 হলে  $\text{OH}^-$  আয়নের ঘনমাত্রা কত?  
●  $1 \times 10^{-2} \text{ M}$   
(খ)  $1 \times 10^{-10} \text{ M}$   
(গ)  $1 \times 10^{-12} \text{ M}$   
(ঘ)  $1 \times 10^{-14} \text{ M}$

- ১৬। 3d অরবিটালের জন্য সহকারী কোয়ান্টাম সংখ্যার মান কত?  
(ক) 0 (খ) 1  
● 2 (গ) 3

- ১৭।  $\text{C}_{12}\text{H}_{22}\text{O}_{11} + \text{H}_2\text{O} \xrightarrow[37^\circ\text{C}]{\text{X}} \text{গ্লুকোজ} + \text{ফ্রুক্টোজ}$   
এই বিক্রিয়ার X হলো—  
(ক) জাইমোস  
● ইনভারটাস  
(গ) অ্যাসিটো ব্যাকটের  
(ঘ) মাইকোভার্ম অ্যাসিট

- ১৮। তাপমাত্রা বিক্রিয়া হলো—  
I.  $\text{X} + \text{Y} + \text{তাপ} \longrightarrow \text{উৎপাদ}$   
II.  $\text{R} + \text{Z} \longrightarrow \text{উৎপাদ}; \Delta H = + V \text{ e}$   
III.  $\text{L} + \text{T} \longrightarrow \text{উৎপাদ} + \text{তাপ}$   
নিচের কোনটি সঠিক?  
● I ও II  
(খ) II ও III  
(গ) I ও III  
(ঘ) I, II ও III

- ১৯। কোন জোড়ের মধ্যে আকৃতি বৈসাম্যুশ বিদ্যমান?  
(ক)  $\text{BF}_3$  ও  $\text{SO}_3$   
(খ)  $\text{H}_2\text{O}$  ও  $\text{H}_2\text{S}$   
(গ)  $\text{POCl}_3$  ও  $\text{BF}_4^-$   
●  $\text{SO}_2$  ও  $\text{CO}_2$

- ২০। বিতর্পার গ্রন্থক ( $R_{90}$ ) এর মান কত?  
(ক)  $1.09678 \times 10^{-2} \text{ m}^{-1}$   
(খ)  $1.09678 \times 10^3 \text{ m}^{-1}$   
(গ)  $1.09678 \times 10^6 \text{ m}^{-1}$   
●  $1.09678 \times 10^7 \text{ m}^{-1}$

- ২১। নাইট্রোজেনযুক্ত সার হলো—  
I. ইউরিয়া  
II. মিউরেট অব পটাস  
III. অ্যামোনিয়াম সালফেট

- নিচের কোনটি সঠিক?  
(ক) I ও II (খ) II ও III  
● I ও III (ঘ) I, II ও III

- ২২। কোন মৌল জোড়া পর্যায় সারণির একই পর্যায়ভুক্ত?  
(ক) Li, Na  
● Mn, Fe  
(গ) Kr, Xe  
(ঘ) B, Al

দ্রবণ উত্তীর্ণক হতে ২৩ ও ২৪নং প্রশ্নের উত্তর দাও :  
একটি কন্ডায়ে 2.5 মোল HI কে  $400^\circ\text{C}$  তাপমাত্রায় উত্তপ্ত করা হল। সাম্যাবস্থায় 25% HI বিয়োজিত হয়।  
 $2 \text{HI(g)} \rightleftharpoons \text{H}_2(\text{g}) + \text{I}_2(\text{g}); \Delta H = - V \text{ e}$

- ২৩। উত্তীর্ণকের বিক্রিয়াক্রম  $K_p$  এর মান কত?  
(ক) 0.16  
● 0.0277  
(গ) 0.0177  
(ঘ) 0.0123

- ২৪। উত্তীর্ণকে বিক্রিয়াক্রম কেড়ে—  
I. তাপমাত্রা বৃদ্ধি পেলে সাম্যের অবস্থান বামদিকে সরে যায়  
II. তাপ বাড়লে  $K_p$  এর মান বাড়ে  
III.  $K_p$  ও  $K_c$  এর মান সমান

- নিচের কোনটি সঠিক?  
(ক) I ও II (খ) II ও III  
● I ও III (ঘ) I, II ও III

- ২৫।  $\text{Ag}_2\text{CrO}_4$  এর প্রত্যাক্তা 'S' হলে প্রত্যাক্তা ৩নংক কত হবে?  
(ক)  $s^2$  ●  $4s^3$   
(গ)  $27s^4$  (ঘ)  $108s^5$

# Hsc Chemistry 1st Paper Mcq Solution 24

**Xiang Xie**



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