

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পরীক্ষার প্রয়োজক

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

সরকের ভর, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = 1$

উদাহরণে উল্লিখিত N_2 -সংশ্লিষ্টে 10 mg এর মধ্যে কয় পরিমাণ N_2 V_1 -সংশ্লিষ্টে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl নমুনা নিয়ে কাজ করা হয়েছে।

যদিও N_2 -সংশ্লিষ্ট হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V_1 -সংশ্লিষ্ট হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায়শই রাসায়নিক উপাদান সন্নিবেশন ও পরিমাপের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।



ফলে, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।

ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

চিত্র-১০



১. উদাহার চুলক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকটির জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহারের অংশের চলক থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার চলক তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ নং প্রশ্নের উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি সারকের জন্য যে ভর পাওয়া যায়, তাকে রাইডার চুলক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায়শই রাসায়নিক উপাদান সন্নিবেশন ও পরিমাপের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৭. উদাহারের চলকটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M চলক। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M চলক হলে পাত্র H_2SO_4 চলক। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্গার প্রতীক অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান চলকে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে মুদ্রণের সময় সংশ্লিষ্ট তরল ও অন্যান্য পদার্থের হ্যাঙ্গারের ক্ষতি সাধিত হয়। এতে তরল পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ (HCl, HNO ₃ , H ₂ SO ₄), পাত্র ক্ষয় (NaOH, KOH), H ₂ O ₂ , অক্সিজেন লিকার প্রভৃতি চলক ইত্যাদি।
সংরক্ষণ পদ্ধতি	<ul style="list-style-type: none"> পাত্র এসিডকে পোতলে রাখতে হবে। পাত্র ক্ষয়কে, ক্ষয়ক পদার্থের পিছনকে পোতলে রাখতে হবে।

৮. উদাহারের চলকটি হলো 10M H_2SO_4 চলক এবং এর থেকে 1M ঘনমাত্রার চলক প্রস্তুত করতে হলে 1ম ও ২য় চলকের ব্যবহার করতে হবে।

1ম চলকটির ঘনমাত্রা 100 ml অক্সিডেন্টিক চলক ও এর টি হলো পিপিটি। সেমি-মাইক্রো অক্সিডেন্টিক চলকের আয়তন 100 ml। তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 চলক তৈরি করতে হবে। চলকের গণনা করা খুব সহজ।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 চলকের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, পিপিটির সাহায্যে 10 ml প্রথম ঘনমাত্রার চলক নিয়ে তার সাথে অক্সিডেন্টিক চলকে 90 ml পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার চলক পাওয়া যাবে।

চিত্র-১১



১. অক্সিডেন্টিক অঙ্কন করে।
২. সামান্য চুলক K₂ এর মান কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিলের হ্যাঙ্গারকে উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিলের সাথে সম্পর্কিত রাসায়নিক দ্রব্যটির ব্যবহারকালে খান্সা কৃতি হলে কবীর উপর সম্পর্ক বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

৫. অক্সিডেন্টিক এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সংরক্ষক যা মুদ্রণ বা অন্য উপাদানসমূহকে জারণ প্রতিরোধ করে।

৬. সামান্য K₂ এর মান কখনো অসীম বা শূন্য হতে পারে না। উল্লিখিত ব্যাখ্যা করার জন্য নিম্নের উত্তরসূরী মাইক্রোপ্রতি বিশ্লেষণ করি—

$$A + B \rightarrow C + D$$

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Xiaolong Qi



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