# Naming Ionic Compounds - Answer Key

Give the name of the following ionic compounds:

#### Name

| 1)  | Na <sub>2</sub> CO <sub>3</sub>   | sodium carbonate                  |
|-----|-----------------------------------|-----------------------------------|
| 2)  | NaOH                              | sodium hydroxide                  |
| 3)  | MgBr <sub>2</sub>                 | magnesium bromide                 |
| 4)  | KCI                               | potassium chloride                |
| 5)  | FeCl <sub>2</sub>                 | iron (II) chloride                |
| 6)  | FeCl <sub>3</sub>                 | iron (III) chloride               |
| 7)  | Zn(OH) <sub>2</sub>               | zinc hydroxide                    |
| 8)  | Be <sub>2</sub> SO <sub>4</sub>   | beryllium sulfate                 |
| 9)  | CrF <sub>2</sub>                  | chromium (II) fluoride            |
| 10) | $AJ_2S_3$                         | aluminum sulfide                  |
| 11) | PbO                               | lead (II) oxide                   |
| 12) | Li <sub>3</sub> PO <sub>4</sub>   | lithium phosphate                 |
| 13) | Titla                             | titanium (IV) iodide              |
| 14) | Co <sub>3</sub> N <sub>2</sub>    | cobalt (II) nitride               |
| 15) | Mg <sub>3</sub> P <sub>2</sub>    | magnesium phosphide               |
| 16) | Ga(NO <sub>2</sub> ) <sub>3</sub> | gallium nitrite                   |
| 17) | Ag <sub>2</sub> SO <sub>3</sub>   | silver sulfite                    |
| 18) | NH <sub>4</sub> OH                | ammonium hydroxide                |
| 19) | AI(CN) <sub>3</sub>               | aluminum cyanide                  |
| 20) | Be(CH <sub>3</sub> CO             | O) <sub>2</sub> beryllium acetate |
|     |                                   |                                   |

# **Formulas Of Ionic Compounds Answer Key**

**Kathleen Armour** 

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