

#### GRADE 9 NATURAL SCIENCES JUNE TEST

SECT							
	6 000	s options are provided as answers rect answer and write ONLY the L TON NUMBER (1.1.1-1.1.10) on y	ETTER	(A-D) next to the			
1.1.1	Which of the following can you NOT learn from a chemical formula?						
	A	Elements in the compound	6	Phase of the element			
	C	Ratio of atoms	D	Total number of atoms			
1.1.2	Which of the following is NOT a sub-atomic particle?						
	A	Electrons	. 8	Protons			
	C	Neutrons	0	Nucleons			
1.1.3	What is the prefix for 6 atoms?						
	A	Hepta-	8	Hexa-			
	C	Pero-	0	Tri-			
1.1.4	An element has an atomic number of 30 and mass number of 65. How many nucleons will if have?						
	A	30	8	35			
	C	65	0	95			
1.1.5	Element at group 15, period 2 on the periodic table						
	A	Carbon	8	Nitragen			
	100	Silicon		Oxygen			

	A	Silicon		Lithium					
	C	Magnesium	0	Carbon					
0.7	Subs	stances on the left side of a chemic	supe los	don:					
	A	Products		Reactants					
	C.	Metals	0	Non-metals					
13.2	Which	it of the following is NOT a way to	preven	trus?					
	A	Galvanisation		Electrolysis					
	¢	Paint	0	Oil or grease					
LER	Sub	stance with a pH lower than 7 will	be diss	sified as:					
	A	Lemon Juice		Neutral					
	c	Apdc	0	Akaine					
1.1.10	An	neutral substance would turn	with	universal indicator.					
	A	Green		Volet					
	Ċ	Colouriess	٥	Red	[10x1]				
		e the correct TERM for the following							
2.1	Meas	sure of how strongly acidic or alkal	ine a p	distance is:					
22	Reaction of a substance with oxygen.								
23	Name for group 18 on the periodic table.								
2.4	Process whereby compounds can be broken down using electricity.								
25	H <sub>0</sub> , C	Do. Bro In							
12.6	Nam	e for groups 3 to 12 on the periodi	c table.						
127	Who	n acids and bases react with each	other.		[50]				

1.1.8 The following is an example of a semi-metal:

# Natural Science Grade 9 Exam Paper 201

**Liying Dong** 

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