Acids and Bases

Question 1:		Choose the correct answer for each of the following questions:				
1.1	The element present in all acids					
	A	oxygen	В	carbon o	dioxide	
	C	nitrogen	D	hydre		
1.2	A sample of an acid in a beaker has a pH of 4. An alka.					pH of
	the liquid in the beaker?					
	A	decreases	В	stays the	e same	
	C	Increases	D	neutral		
1.3	Choose the incorrect combination:					
	A	HNH₄OH – Nitric acid.	В	H ₂ SO	sulphuric ac	
	C	HCl - Hydrochloric acid.	D		rbonic acid.	
1.4	The colour of blue litmus paper in an acid is					
	A	yellow	В	en		
	C	blue				
1.5	Which substance is always produced in eaction. ochloric and and lithium?					
	A	water		hydrogen 8-		
	C	oxygen gas		on dioxide gas		
1.6	Which:	substance is always produ in	the rea.		n hydrochloric acid and sodium	
	hydroxi	ide?				
	A	water		hydrogen gas		
	C	oxygen gas		ecipitate		
1.7	Which formula repr					
	A	кон		KCI		
	C	СН₃ОН		CH₃COOH		
1.8	T	od of adding an a	irop b	y drop:		
		neutralisation	В	precipita	ation	
		eva-	D	decomp	osition	[8]
Quest.	.ne genera reaction for an			acid with	the following compounds:	
2.1	Base:					
2.2	Carbonate:					
2.3	Metal:			-(8)-5	turdvehamo-co.	za

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WJ Hussar

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Decoding Natural Science Exam Grade 9 2014 November: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "Natural Science Exam Grade 9 2014 November," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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