

Assignment #3 – Compound Names and Formulas Single and Multi-valent ions

A. Write the correct formula for the following compounds.

1. gallium oxide Ga_2O_3
2. mercury (I) chloride $HgCl$
3. tin (IV) bromide $SnBr_4$
4. indium oxide In_2O_3
5. cadmium sulphide CdS
6. magnesium oxide MgO
7. potassium nitride K_3N
8. iron (III) sulphide Fe_2S_3
9. gold (III) bromide $AuBr_3$
10. zinc oxide ZnO

11. silicon oxide SiO_2
12. aluminum bromide $AlBr_3$
13. tin (II) sulphide SnS
14. germanium bromide $GeBr_4$
15. nickel (II) oxide NiO
16. strontium astatide $SrAt_2$
17. aluminum oxide Al_2O_3
18. copper (II) oxide CuO
19. lawrencium fluoride LrF_3
20. actinium oxide Ac_2O_3

B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.

1. $CaBr_2$ Calcium Bromide
2. K_2O Potassium Oxide
3. Fe_2O_3 Iron (III) Oxide
4. ZnF_2 Zinc Fluoride
5. $CuCl$ Copper (I) Chloride
6. Ni_2O_3 Nickel (III) Oxide
7. $AuCl$ Gold (I) Chloride
8. $HgBr$ Mercury (I) Bromide

9. $AsCl_5$ Arsonic (V) Chloride
10. Eu_2O_3 Europium (III) Oxide
11. $NdBr_3$ Neodymium Bromide
12. Cu_3N_2 Copper (II) Nitride
13. CoS Cobalt (II) Sulphide
14. ZrO_2 Zirconium Oxide
15. Li_2O Lithium Oxide
16. NaH Sodium Hydride

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S Ashworth



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Table of Contents Naming Of Chemical Compounds Worksheets With Answers

1. Understanding the eBook Naming Of Chemical Compounds Worksheets With Answers
 - The Rise of Digital Reading Naming Of Chemical Compounds Worksheets With Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Of Chemical Compounds Worksheets With Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Of Chemical Compounds Worksheets With Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Of Chemical Compounds Worksheets With Answers
 - Personalized Recommendations
 - Naming Of Chemical Compounds Worksheets With Answers User Reviews and Ratings

- Naming Of Chemical Compounds Worksheets With Answers and Bestseller Lists
- 5. Accessing Naming Of Chemical Compounds Worksheets With Answers Free and Paid eBooks
 - Naming Of Chemical Compounds Worksheets With Answers Public Domain eBooks
 - Naming Of Chemical Compounds Worksheets With Answers eBook Subscription Services
 - Naming Of Chemical Compounds Worksheets With Answers Budget-Friendly Options
- 6. Navigating Naming Of Chemical Compounds Worksheets With Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Of Chemical Compounds Worksheets With Answers Compatibility with Devices
 - Naming Of Chemical Compounds Worksheets With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Of Chemical Compounds Worksheets With Answers
 - Highlighting and Note-Taking Naming Of Chemical Compounds Worksheets With Answers
 - Interactive Elements Naming Of Chemical Compounds Worksheets With Answers
- 8. Staying Engaged with Naming Of Chemical Compounds Worksheets With Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Of Chemical Compounds Worksheets With Answers
- 9. Balancing eBooks and Physical Books Naming Of Chemical Compounds Worksheets With Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Of Chemical Compounds Worksheets With Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Of Chemical Compounds Worksheets With Answers
 - Setting Reading Goals Naming Of Chemical Compounds Worksheets With Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Of Chemical Compounds Worksheets With Answers
 - Fact-Checking eBook Content of Naming Of Chemical Compounds Worksheets With Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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