

Naming Compounds Containing Polyatomic Ions

Use the following list of polyatomic ions to name the compounds given.

OH
NO₃
C₂H₃O₂

hydroxide
nitrate
acetate

ClO₃
HCO₃
SO₄
CO₃

chlorate
bicarbonate
sulfate
carbonate

SO₃
PO₄
NH₄

sulfite
phosphate
ammonium

- | | |
|--|---|
| 1. CuSO ₄ _____ | 26. NaOH _____ |
| 2. KClO ₃ _____ | 27. NaNO ₃ _____ |
| 3. NH ₄ OH _____ | 28. Ba(OH) ₂ _____ |
| 4. K ₂ CO ₃ _____ | 29. K ₂ CO ₃ _____ |
| 5. Na ₂ SO ₄ _____ | 30. LiOH _____ |
| 6. KC ₂ H ₃ O ₂ _____ | 31. KHCO ₃ _____ |
| 7. NH ₄ Cl _____ | 32. NH ₄ F _____ |
| 8. Mg(OH) ₂ _____ | 33. KClO ₃ _____ |
| 9. Zn(NO ₃) ₂ _____ | 34. NaC ₂ H ₃ O ₂ _____ |
| 10. MgSO ₄ _____ | 35. CaClO ₃ _____ |
| 11. CaCO ₃ _____ | 36. SrNO ₃ _____ |
| 12. KOH _____ | 37. Be(OH) ₂ _____ |
| 13. Ca(OH) ₂ _____ | 38. Li ₂ SO ₄ _____ |
| 14. K ₂ CO ₃ _____ | 39. Al ₂ (SO ₄) ₃ _____ |
| 15. AlPO ₄ _____ | 40. BaSO ₃ _____ |
| 16. Na ₂ PO ₄ _____ | 41. NH ₄ I _____ |
| 17. MgSO ₄ _____ | 42. NH ₄ HCO ₃ _____ |
| 18. AgNO ₃ _____ | 43. Cu ₂ (PO ₄) ₃ _____ |
| 19. Na ₂ CO ₃ _____ | 44. Ni(OH) ₂ _____ |
| 20. NaHCO ₃ _____ | 45. ZnSO ₄ _____ |
| 21. Ca(NO ₃) ₂ _____ | 46. MnCO ₃ _____ |
| 22. BaSO ₄ _____ | 47. H ₂ SO ₄ _____ |
| 23. BPO ₃ _____ | 48. Au ₂ SO ₄ _____ |
| 24. LiNO ₃ _____ | 49. NH ₄ NO ₃ _____ |
| 25. Al(NO ₃) ₃ _____ | 50. LiClO ₃ _____ |

Naming Ionic Compounds With Polyatomic Ions Answers

NARAYAN CHANDER



Naming Ionic Compounds With Polyatomic Ions Answers:

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Table of Contents Naming Ionic Compounds With Polyatomic Ions Answers

1. Understanding the eBook Naming Ionic Compounds With Polyatomic Ions Answers
 - The Rise of Digital Reading Naming Ionic Compounds With Polyatomic Ions Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Naming Ionic Compounds With Polyatomic Ions Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Naming Ionic Compounds With Polyatomic Ions Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Naming Ionic Compounds With Polyatomic Ions Answers
 - Personalized Recommendations

- Naming Ionic Compounds With Polyatomic Ions Answers User Reviews and Ratings
- Naming Ionic Compounds With Polyatomic Ions Answers and Bestseller Lists
- 5. Accessing Naming Ionic Compounds With Polyatomic Ions Answers Free and Paid eBooks
 - Naming Ionic Compounds With Polyatomic Ions Answers Public Domain eBooks
 - Naming Ionic Compounds With Polyatomic Ions Answers eBook Subscription Services
 - Naming Ionic Compounds With Polyatomic Ions Answers Budget-Friendly Options
- 6. Navigating Naming Ionic Compounds With Polyatomic Ions Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Naming Ionic Compounds With Polyatomic Ions Answers Compatibility with Devices
 - Naming Ionic Compounds With Polyatomic Ions Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Naming Ionic Compounds With Polyatomic Ions Answers
 - Highlighting and Note-Taking Naming Ionic Compounds With Polyatomic Ions Answers
 - Interactive Elements Naming Ionic Compounds With Polyatomic Ions Answers
- 8. Staying Engaged with Naming Ionic Compounds With Polyatomic Ions Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Naming Ionic Compounds With Polyatomic Ions Answers
- 9. Balancing eBooks and Physical Books Naming Ionic Compounds With Polyatomic Ions Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Naming Ionic Compounds With Polyatomic Ions Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Naming Ionic Compounds With Polyatomic Ions Answers
 - Setting Reading Goals Naming Ionic Compounds With Polyatomic Ions Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Naming Ionic Compounds With Polyatomic Ions Answers
 - Fact-Checking eBook Content of Naming Ionic Compounds With Polyatomic Ions Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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