

Naming Ionic Compounds – Answer Key

Give the name of the following ionic compounds:

		Name
1)	Na_2CO_3	sodium carbonate
2)	NaOH	sodium hydroxide
3)	MgBr_2	magnesium bromide
4)	KCl	potassium chloride
5)	FeCl_2	iron (II) chloride
6)	FeCl_3	iron (III) chloride
7)	Zn(OH)_2	zinc hydroxide
8)	Be_2SO_4	beryllium sulfate
9)	CrF_2	chromium (II) fluoride
10)	Al_2S_3	aluminum sulfide
11)	PbO	lead (II) oxide
12)	Li_3PO_4	lithium phosphate
13)	TiI_4	titanium (IV) iodide
14)	Co_3N_2	cobalt (II) nitride
15)	Mg_3P_2	magnesium phosphide
16)	$\text{Ga(NO}_2)_3$	gallium nitrite
17)	Ag_2SO_3	silver sulfite
18)	NH_4OH	ammonium hydroxide
19)	Al(CN)_3	aluminum cyanide
20)	$\text{Be(CH}_3\text{COO)}_2$	beryllium acetate

Naming Ionic Compounds Key

David Callaghan

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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