

Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being in their relatively different mass. However, when the elements react, combinations formed are consistent in their composition. Why? What does this evidence suggest about the particles that make up the atoms for which there is evidence a tendency to join to form the compounds that we see today?

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic symbols and formulas	12 hydrogen 12 oxygen H_2O_2	12 hydrogen 12 oxygen H_2O	12 hydrogen 12 sulfur H_2S_2	1 hydrogen 1 oxygen HO
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particle



diagram

Atomic symbols and formulas	12 sulfur 12 oxygen SO_2	12 hydrogen 12 oxygen H_2O_2	12 sulfur 12 oxygen SO_2
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particle



diagram

Atomic symbols and formulas	12 hydrogen 12 chlorine HCl	12 hydrogen 12 chlorine H_2Cl_2	1 hydrogen 1 chlorine HCl	1 hydrogen 1 chlorine HCl
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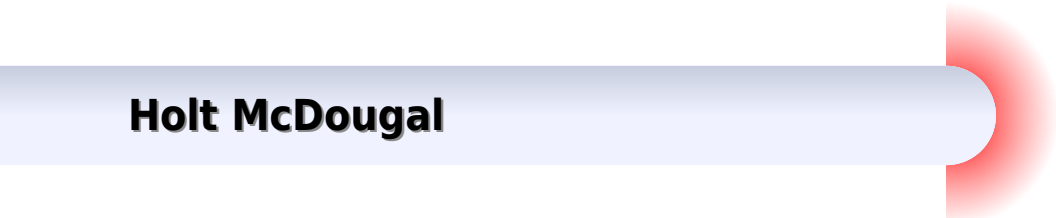
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diagram

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Integrated Mathematics Holt McDougal, 1998 **Glencoe Mathematics**, 2001

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