

# John Manuel

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## MOUSSAKA TO MY EARS

VOLUME TWO  
OF THE "RAMBLINGS FROM RHODES" SERIES



# Moussaka To My Ears Ramblings From Rhodes English Edition

**john manuel**



### **Moussaka To My Ears Ramblings From Rhodes English Edition:**

Moussaka to My Ears john manuel, 2009-05-11 John and his half Greek wife Yvonne Maria moved from the UK to Rhodes in 2005 the account of which was told in his first book FETA COMPLI MOUSSAKA takes up where FETA COMPLI leaves off with Yvonne and John settling into life on Rhodes and coming into contact with some great characters who live around them out there Plus there s another helping of reminiscences from past years and other parts of Greece and Europe in general like the time when Yvonne was Miss August on a calendar that graced the walls of innumerable offices all across Germany after meeting a German businessman on a beach in Corfu Then there s Manolis the Six Million Drachma Man who climbs olive trees even though he only has one and a half legs The Lonely Goatherd comes by as does Kostas the local who shows them which weeds are great to eat Yvonne helps Greco Turkish relations during an evening out on Samos and John reluctantly become the elephant man for a while

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web donor whereas a base generically rep resented by b is a proton acceptor when acids and bases combine they swap a proton ha b a hb strong acids e g sulfuric acid have a very pronounced ten dency to transfer a proton to a base whereas weak acids

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web understand that parts of an acid base titration be able to determine the k a or k b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to



calculate  $K_{a1}$  and  $K_{a2}$  for a polyprotic acid by the end of this lab students should be able to

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web jul 19 2023 at the beginning of the titration the solution in the erlenmeyer flask is acidic as the base is added it completely reacts with the acid and the solution in the erlenmeyer flask continues to be acidic but at the equivalence point [acid base titration chemistry 1210 lab report containing an](#) - Oct 07 2023

web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the neutral pH at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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web 1 approximately 3 g of hydrated oxalic acid  $H_2C_2O_4 \cdot 2H_2O$  was weighted to the nearest 0.01 g in a 50 ml beaker 2 about 30 ml of distilled water was added to the beaker to dissolve the oxalic acid 3 the solution was transferred into 250 ml volumetric flask the beaker was rinsed and its contents were poured into the flask

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web solution a titrant volume 0 ml the solution ph is due to the acid ionization of hcl because this is a strong acid the ionization is complete and the hydronium ion molarity is 0 100 m the ph of the solution is then  $\text{ph} = -\log 0.100 = 1.000$  b titrant volume 12.50 ml

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web one type of titration uses a neutralization reaction in which an acid and a base react to produce a salt and water in equation 1 the acid is hcl hydrochloric acid and the base is naoh sodium hydroxide when the acid and base react they form nacl sodium chloride which is also known as table salt

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