

### Experiment 5-Worksheet Nomenclature

Write formula or name for each compound:

#### IA. Binary ionic compounds

	Name	Formula	Formula	Name
1	sodium oxide	$\text{Na}_2\text{O}$	$\text{BaS}$	
2	calcium nitride		$\text{AlBr}_3$	
3	barium iodide		$\text{Ca}_3\text{P}_2$	
4	potassium sulfide		$\text{Mg}_3\text{N}_2$	
5	aluminum oxide		$\text{K}_2\text{O}$	

#### IB. Transition metals with one oxidation state

	Name	Formula	Formula	Name
1	silver oxide		$\text{ZnCl}_2$	
2	zinc phosphide		$\text{AgI}$	
3	zinc sulfide		$\text{Ag}_3\text{P}$	

#### IC. Metals more than one oxidation state

	Name	Formula	Formula	Name
1	cupric oxide		$\text{FeBr}_3$	
2	iron (III) sulfide		$\text{CuI}$	
3	ferrous nitride		$\text{CuS}$	

# Formulas And Nomenclature Oxidation Lab

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*Laboratory Directions and Study Questions in Inorganic Chemistry* Alexander Silverman, 1919

**Merrill Laboratory Chemistry** L. Neal Carmichael, David F. Haines, 1990-10

Laboratory Manual Jo A. Beran, 1990

**A Practical Elementary Chemistry** Boynton Wells

McFarland, 1915

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Catalogue George Washington University, 1903

**Announcement** George Washington University. Dept. of Arts and Sciences, 1907

**Chemistry Workbook and Laboratory Guide** Martin Van Buren McGill, 1933

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*Bulletin and Catalog* Muskegon County Community College, New Concord Muskingum college (Ohio), 1961

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