

# NAMING COMPOUNDS: PRACTICE

Name the following covalent compounds:

SO<sub>3</sub>      sulfur trioxide

N<sub>2</sub>S      dinitrogen <sup>mono</sup>sulfide

P<sub>2</sub>Br<sub>4</sub>      diphosphorus tetrabromide

CO      carbon monoxide (not ~~monoxide~~)

NO<sub>2</sub>      nitrogen dioxide

# Formula Writing For Covalent Compounds

**Graeme Lofts, Merrin J. Evergreen**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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*Chemistry* Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been

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**Introductory Basics Of Chemistry** Dr J. Sai Chandra,2022-05-07 Our Chemistry Reference Book adheres to the scope and sequence of most general chemistry courses nationwide We strive to make chemistry as a discipline interesting and accessible to students With this objective in mind the content of this Reference Book has been developed and arranged to provide a logical progression from fundamental to more

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