CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

Candidates are allowed additional 15 minutes for only reading the paper.

They must NOT start writing during this time.

This paper is divided into four sections - A. B. C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B, Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets []. Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant R = 1.987 cal deg' mot' = 8.314 JK' mot' = 0.0821 dm' atm K'mot'

I I atm = I dm3 atm = 1013 J. I Faraday = 96500 coulombs.

Avogadro's number = 6.023×10^{19} .

SECTION A – 14 MARKS

Ouestion 1

(A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1] brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into _____ and then into
- (ii) The intermediate activated complex in a chemical reaction is highly _____ due to _____ energy.
- (iii) The coordination number and exidation state of the complex K₄[Fe(CN)₆] are _____ and ____ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives and the reaction is known as _____ reduction.

This Paper consists of 8 printed pages.

Turn over

Industrial Chemistry Sample Question Paper

RD Boyd

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Table of Contents Industrial Chemistry Sample Question Paper

- 1. Understanding the eBook Industrial Chemistry Sample Question Paper
 - The Rise of Digital Reading Industrial Chemistry Sample Question Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Industrial Chemistry Sample Question Paper
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Industrial Chemistry Sample Question Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Industrial Chemistry Sample Question Paper
 - Personalized Recommendations
 - Industrial Chemistry Sample Question Paper User Reviews and Ratings

- Industrial Chemistry Sample Question Paper and Bestseller Lists
- 5. Accessing Industrial Chemistry Sample Question Paper Free and Paid eBooks
 - Industrial Chemistry Sample Question Paper Public Domain eBooks
 - Industrial Chemistry Sample Question Paper eBook Subscription Services
 - Industrial Chemistry Sample Question Paper Budget-Friendly Options
- 6. Navigating Industrial Chemistry Sample Question Paper eBook Formats
 - o ePub, PDF, MOBI, and More
 - Industrial Chemistry Sample Question Paper Compatibility with Devices
 - Industrial Chemistry Sample Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Industrial Chemistry Sample Question Paper
 - Highlighting and Note-Taking Industrial Chemistry Sample Question Paper
 - Interactive Elements Industrial Chemistry Sample Question Paper
- 8. Staying Engaged with Industrial Chemistry Sample Question Paper
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Industrial Chemistry Sample Question Paper
- 9. Balancing eBooks and Physical Books Industrial Chemistry Sample Question Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Industrial Chemistry Sample Question Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Industrial Chemistry Sample Question Paper
 - Setting Reading Goals Industrial Chemistry Sample Question Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Industrial Chemistry Sample Question Paper
 - Fact-Checking eBook Content of Industrial Chemistry Sample Question Paper
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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neither

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