Types of Reactions Worksheet

Balance the following equations and indicate the type of reaction taking place:

1) _____NaBr + _____HaPO4 → _____NaaPO4 + ____HBr

Type of reaction:

2) ____ Ca (OH)₂ + ____ Al₂(SO₄)₃ → ____ CaSO₄ + ____ Al(OH)₃

Type of reaction:

3) ____ Mg + ___ Fe₂O₃ → ___ Fe + ___ MgO

Type of reaction: _____

4) C₂H₄ + ____ O₂ → ____ CO₂ + ____ H₂O

Type of reaction:

5) ____ PbSO₄ → ____ PbSO₃ + ____ O₂

Type of reaction: ____

6) ____NH₃ + ____N₂I₆ + ____H₂

Type of reaction:

7) ____ H₂O + ____ SO₃ → ____ H₂SO₄

Type of reaction: _____

8) _____ $H_2SO_4 +$ ____ $NH_4OH \rightarrow$ _____ $H_2O +$ ____ $(NH_4)_2SO_4$

Type of reaction:

Indicate The Type Of Reaction Taking Place

M Mark

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