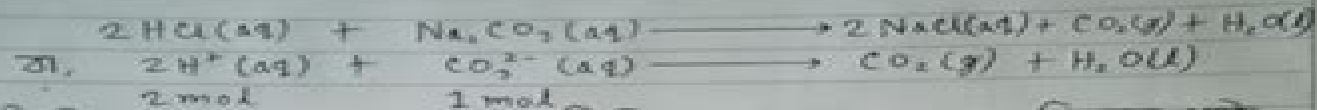


NAME OF THE EXPERIMENT: 0.1M Na_2CO_3	Page No.	
উদ্যম দ্বারা সংকটগ্রস্ত নমুনা উত্তোলন (HCl উত্তোলন)	Expt No.	01
চলমান নিন্দিত	Date :	
মূলনীতি:		

Na_2CO_3 ও HCl এর প্রকৃষ্ট বিক্রিয়া সমীকরণ নিম্নরূপ:



টাইট্রেশনের তৈরিতে, প্রকৃষ্ট বিক্রিয়া কোষ প্রকৃষ্ট বিক্রিয়ায় পরিণত, CO_3^{2-} এর মোল সংখ্যা (ফ্রাক্ট) $\equiv \text{H}^+$ এর মোল সংখ্যা (ক্যালি ব্রোমিডে দেয়া)

প্রয়োজনীয় সামগ্র্যনিক বস্তু:

- 1) 0.1M Na_2CO_3 দ্রব
- 2) 0.1M HCl দ্রব
- 3) নির্দেশক : মিথাইল অরেঞ্জ

প্রয়োজনীয় যন্ত্রপাতি:

- 1) কনিক্যাল ফ্লাস্ক
- 2) ব্যালো
- 3) পিপেট
- 4) ক্যাম্পমহ মেসার

কার্যের ধারা:

- 1) প্রস্তুত করা 0.1M HCl দ্রব (প্রকৃষ্টাতি অম্ল দ্রব) অংশ নমুনা HCl দ্বারা ব্যালোটি বর্ণ করি এবং মেসারের মাধ্যমে ক্যাম্পমহ দ্বারা পরিমাপ করে আটকিয়ে নিই।
- 2) প্রস্তুত করা 0.1M Na_2CO_3 দ্রব (প্রকৃষ্ট দ্রব) এর আনুমানিক ফ্রাক্টের মধ্যে পিপেট দ্বারা 10ml দ্রব কনিক্যাল ফ্লাস্কে নিই, কনিক্যাল ফ্লাস্কে দ্রব 1-2 ফোটা মিথাইল অরেঞ্জ যোগ করি, তখন দ্রবের বর্ণ হলুদ হয়।
- 3) কনিক্যাল ফ্লাস্কে তৈরি একটি সাদা কাগজ টুকরা রাখি, ব্যালোটি HCl অম্ল দ্রবের দিকে নিয়ে ঘোঁরা দিই 'গম পাঠ' - এ ফেরি করি, চিহ্নযুক্ত সাদা সাদা ব্যালোটির দৈর্ঘ্য লক্ষ্য করে ব্যালোটির HCl দ্রব ফোঁটা ফোঁটা কনিক্যাল ফ্লাস্কে যোগ করি

Hsc Chemistry 2nd Paper Practical Book 23

Rosina Ehmann



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- Chapter 5: Conclusion

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