

i.) 207.35

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 1

Q1. At the same temperature and pressure, an equal volume of gases contains the same number
of
a.) Molecules
b.) Electrons
c.) Protons
d.) Particles
Q2. The gas law that relates the volume of a gas to the number of molecules of the gas is
a.) Gay-Lussac's law
b.) Avogadro's law
c.) Boyle's law
d.) Charle's law
Q3. The mass % of carbon in CO ₂ is
a.) 0.034%
b.) 27.27%
c.) 3.4%
d.) 28.7%
Q4. The empirical formula of benzene is
a.) CH
b.) C _o H _e
c.) C ₂ H ₃
d.) C ₂ H ₂
Q5. What is the mole fraction of the solute in a 1.00 molal aqueous solution?
a.) 0.00177
b.) 0.0344
c.) 0.0177
d.) 0.1770

Q6. State the number of significant figures in each of the following numbers:

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Wenbin Ji

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Holt Chemistry Ralph Thomas Myers, 2004 & Winston, Holt, Rinehart and Winston Staff, 2001 Winston, Holt, Rinehart and Winston Staff, 2004 Winston, Holt, Rinehart and Winston Staff, 2004 Winston, 1998-02

HRW Algebra One Interactions, 2001 Modern Chemistry Holt Rinehart Chapter Resource 1 Biology and You Biology Holt Rinehart & Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Videodisc Correlatn GD Modern Biology 99 Holt Rinehart & Elements of Experimental Chemistry René Lahaie, Luc Papillon, Pierre Valiquette, 1978

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Progress in Medicinal Chemistry, 1996-04-24 Progress in Medicinal Chemistry

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