

Empirical & Molecular Formulas

Caffeine ($\text{C}_8\text{H}_{10}\text{N}_4\text{O}_2$) is found in tea and coffee .
Write its empirical formula ?

Given:

Molecular formula = $\text{C}_8\text{H}_{10}\text{N}_4\text{O}_2$

Find out its empirical formula

Solution:

Empirical formula = $\text{C}_4\text{H}_5\text{N}_2\text{O}$

$$\begin{array}{cccc} \text{C}_8 & \text{H}_{10} & \text{N}_4 & \text{O}_2 \\ \hline \cancel{8} & : & \cancel{10} & : & \cancel{4} & : & \cancel{2} \\ \hline \cancel{2} & & \cancel{2} & & \cancel{2} & & \cancel{2} \\ & & 4 & : & 5 & : & 2 & : & 1 \end{array}$$

Important Questions for Exams

How To Calculate Empirical And Molecular Formula Books

Shasha Hu



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web answer d somewhere in region d justification with the test charge and c 1 being negative there is a repulsive force on the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb s law $\frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r^2}$ we know that the force

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