

PACING GUIDE FOR Chemistry - Grade 11 or 12

MONTHS	CYCLES	Standards (*State standards; others for alignment with Comm. Coll. Curriculum; essential skills are bold)	Glencoe Chemistry Matter and Change	Resources For Differentiation
August / September	1	Safety Intro States of matter Physical and chemical properties and changes	1.4 (p. 14-17) and lab 1.1 (to demonstrate safe lab practices) 1.2 (p. 7-9) 3.1(p. 54-60) 3.2(p.61-65) Lab 3.1 (p.17) Lab 3.2 (p.21) MiniLab (p.68) Demo – gold pennies p. 62- 63 Small-scale lab 2 Forensics lab A	ACS safety video, worksheets, and quiz Lab - Observing physical and chemical changes Lab – Density of soda
September/ October	2	Atomic structure • "The nucleus of the atom is much smaller than the atom yet contains most of its mass. • "The quantum model of the atom is based on experiments and analyses by many scientists, including Dalton, Thomson, Bohr, Rutherford, Milliken, and Einstein. • "The position of an element in the periodic table is related to its atomic number.	4.1 4.2 4.3 MiniLab p. 102 – isotopes (pennies) Lab 4.1 (p. 25) Lab p. 108-109 (very small particles) Discovery Lab p. 117 (or black box lab)	"Black box" Lab

High School Chemistry Pacing Guide

Shasha Hu



High School Chemistry Pacing Guide:

High School Chemistry Pacing Guide Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**High School Chemistry Pacing Guide**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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