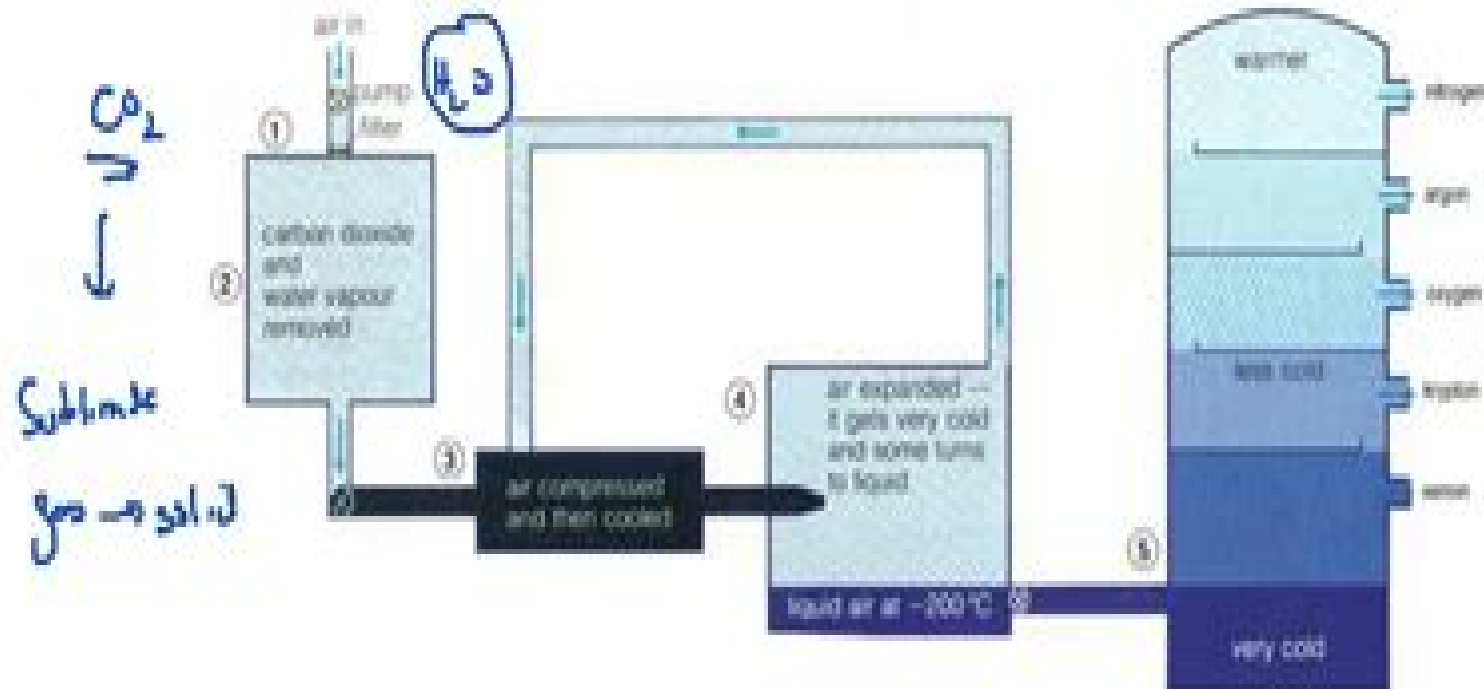


(Extended only) Describe the separation of oxygen and nitrogen from liquid air by fractional distillation

- Fractional distillation of liquid air separates oxygen and nitrogen
- oxygen has a boiling point higher than that of nitrogen
- air is liquefied (around -200°C)
- fractional distillation column temperature is increased until nitrogen boils and rises to the top as a gas (around -196°C)
- the nitrogen is collected and removed
- the temperature is increased until oxygen boils and rises to the top as a gas (around -183°C) and oxygen is collected



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