

## F324-unofficial mark scheme (in progress)

# Marks accounted for: 60/60

- Give the systematic name of the alcohol that is formed by this hydrolysis. (1)
  - Propan(e)-1,2,3-triol
- What are the other monomers that make up this polymer? (3)  
(Hydrolysed by hot NaOH)
  - Two different carboxylic acids with  $\text{COO}^-(\text{Na}^+)$  - The monomers have  $\text{COOH}$  as it's not referring to the hydrolysed products but the monomer units of the polymer I'm pretty sure that the question said: "What are the other products of this hydrolysis" or something along those lines
  - One salt had a  $\text{C}=\text{C}$  double bond
- Why are people who drink a lot of goat's milk likely to suffer from CHD? (2)
  - Fatty acids build up in the arteries and lead to CHD?
  - Has trans isomer due to  $\text{C}=\text{C}$  double bond fatty acid linked to LDL's and hence Atherosclerosis
  - Trans-fatty acid in triglyceride of milk increases 'bad' cholesterol levels in blood, resulting in increased risk of coronary heart disease and stroke (straight from the spec)
  - trans fatty acids can pack more tightly and hence have higher melting points than cis fatty acids.  $\rightarrow$  so more likely to deposit on artery walls
  - The other two molecules were fully saturated fatty acids which would also lead to the formation of atheroma
- State how 4-aminophenol can behave as a base. (1)
- The lone pair on the Nitrogen atom in the amine group can accept a proton, forming a dative covalent bond with it to give  $\text{NH}_3^+$ .
- $\text{NH}_2 + \text{H}^+ \rightarrow \text{NH}_3^+$
- 4-Aminophenol is produced by the reduction of 4-nitrophenol. Write an equation to show the production of 4-aminophenol from 4-nitrophenol. Use  $[\text{H}]$  to represent the reducing agent. (1)
- $4\text{-Nitrophenol} + 6[\text{H}] = 4\text{-Aminophenol} + 2\text{H}_2\text{O}$
- 4-nitrophenol can be produced from 4-bromophenol. (4)
  - Complete the mechanism for this reaction.
  - Use  $^+\text{NO}_2$  as the electrophile. Include any intermediate and the products.
  - In the mechanism for this reaction,  $\text{NO}_2$  substitutes for Br on the ring.

# F324 June 2014 Unofficial Mark Scheme

**LM Reder**



**F324 June 2014 Unofficial Mark Scheme:**

The book delves into F324 June 2014 Unofficial Mark Scheme. F324 June 2014 Unofficial Mark Scheme is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into F324 June 2014 Unofficial Mark Scheme, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to F324 June 2014 Unofficial Mark Scheme
    - Chapter 2: Essential Elements of F324 June 2014 Unofficial Mark Scheme
    - Chapter 3: F324 June 2014 Unofficial Mark Scheme in Everyday Life
    - Chapter 4: F324 June 2014 Unofficial Mark Scheme in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, the author will provide an overview of F324 June 2014 Unofficial Mark Scheme. This chapter will explore what F324 June 2014 Unofficial Mark Scheme is, why F324 June 2014 Unofficial Mark Scheme is vital, and how to effectively learn about F324 June 2014 Unofficial Mark Scheme.
  3. In chapter 2, the author will delve into the foundational concepts of F324 June 2014 Unofficial Mark Scheme. The second chapter will elucidate the essential principles that must be understood to grasp F324 June 2014 Unofficial Mark Scheme in its entirety.
  4. In chapter 3, this book will examine the practical applications of F324 June 2014 Unofficial Mark Scheme in daily life. The third chapter will showcase real-world examples of how F324 June 2014 Unofficial Mark Scheme can be effectively utilized in everyday scenarios.
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  6. In chapter 5, the author will draw a conclusion about F324 June 2014 Unofficial Mark Scheme. This chapter will summarize the key points that have been discussed throughout the book.
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