

**Purpose:** To study the ideal gas law and apply the first law of thermodynamics to analyze behavior of gas. Go to <http://phet.colorado.edu/en/simulation/gas-properties> and run the simulation

**Procedure****PART A: THE NATURE OF THE IDEAL GAS LAW**

1: What happens to the pressure in the chamber if you heat the gas?

2: What happens to the temperature in the chamber if you compress the gas?

3: Locate the "Constant Parameter" section of the on-screen control panel. Lock the temperature. Use "Scubie" to slowly compress the gas. What happens to temperature, and what action is taken (by the simulation) to maintain constant temperature?

4: Lock the pressure. (Doing so releases the lock on temperature.) Add heat. What happens to pressure, and what action is taken (by the simulation) to maintain constant pressure?

**PART B: Kinetic Molecular Theory and the First Law of Thermodynamics**

For each of the following four steps record the average pressure (in atm) and the volume

**Table of Pressure and Volume for each step**

Step	1	2	3	4
Pressure (atm)				
Volume ( $\text{m}^3$ )				

1. Add 300 molecules of the heavy species to the chamber and let the temperature and pressure come to equilibrium.
  - a. Using Avogadro's number express the amount of gas in terms of moles.
  - b. Calculate the internal energy ( $U$ ) of the heavy oxygen gas in joules .
  - c. The container is rectangular in shape with an area of  $2.81 \times 10^{-17} \text{ m}^2$ . Using the ruler tool (in measurement tools) measure the interior length of the container and report measurement in meters.  
Length: \_\_\_\_\_ m
  - d. Calculate the volume of the container (in  $\text{m}^3$ )
  - e. Create different volumes and record the average pressure (in atm) and the volume in the table above for step 1.

# Exploring Gas Laws Lab Answers Phet

**Judd E. Hollander**



## **Exploring Gas Laws Lab Answers Phet:**

## Whispering the Secrets of Language: An Mental Quest through **Exploring Gas Laws Lab Answers Phet**

In a digitally-driven world wherever monitors reign supreme and immediate transmission drowns out the subtleties of language, the profound secrets and emotional subtleties concealed within phrases often move unheard. Yet, nestled within the pages of **Exploring Gas Laws Lab Answers Phet** a captivating literary value pulsating with organic thoughts, lies an extraordinary quest waiting to be undertaken. Written by an experienced wordsmith, that marvelous opus invites viewers on an introspective journey, gently unraveling the veiled truths and profound affect resonating within the very cloth of every word. Within the psychological depths of this moving evaluation, we will embark upon a sincere exploration of the book is primary themes, dissect its interesting writing type, and succumb to the effective resonance it evokes deep within the recesses of readers hearts.

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