
Quiz

Section: Exploring the Periodic Table

In the space provided, write the letter of the answer choice that best completes each statement or best answers each question.

- _____ 1. Helium is the heaviest of elements in the periodic table, as measured
 - a. color.
 - b. number of electrons.
 - c. number of protons.
 - d. atomic weight.
- _____ 2. Electronegativity refers to the tendency for an atom to attract
 - a. valence electrons.
 - b. positive electrons.
 - c. atomic electrons.
 - d. negative electrons.
- _____ 3. Because the pattern is consistent, all groups of elements react in the same way with chlorine. Which of the following is not a group?
 - a. noble gases.
 - b. alkali metals.
 - c. transition metals.
 - d. rare earths.
- _____ 4. A group that acts as a base is known as a
 - a. metal.
 - b. nonmetal.
 - c. halogen.
 - d. noble gas.
- _____ 5. Electronegativity refers to the tendency for an atom to
 - a. negatively attract.
 - b. attract electrons.
 - c. positively attract.
 - d. attract other atoms.
- _____ 6. Group 17 elements have
 - a. metals.
 - b. nonmetals.
 - c. metalloids.
 - d. semimetals.
- _____ 7. The three main groups of elements are metals, metalloids, and
 - a. noble gases.
 - b. alkali metals.
 - c. noble gases.
 - d. semimetals.
- _____ 8. Electronegativity
 - a. metals.
 - b. metalloids.
 - c. nonmetals.
 - d. semimetals.
- _____ 9. Electronegativity is the pull on the valence electrons and
 - a. noble gases.
 - b. metals.
 - c. metalloids.
 - d. semimetals.
- _____ 10. Electronegativity
 - a. metals.
 - b. metalloids.
 - c. nonmetals.
 - d. semimetals.

Exploring The Periodic Table Quiz Holt Science Spectrum Answers

George Graybill



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