Name: Date:
-------------

#### Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

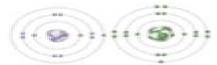
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

#### Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lanic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (CI) are selected from the menus at right. Click Play ( - to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



 Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons are called valence electrons.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

Click Pause ( " ). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

# **Explore Learning Ionic Bonds Answers**

**Sebastian Brünink** 



Whispering the Techniques of Language: An Emotional Journey through Explore Learning Ionic Bonds Answers

In a digitally-driven earth where monitors reign supreme and immediate conversation drowns out the subtleties of language, the profound secrets and psychological subtleties hidden within phrases usually go unheard. Yet, nestled within the pages of **Explore Learning Ionic Bonds Answers** a fascinating literary treasure pulsing with fresh feelings, lies an exceptional journey waiting to be undertaken. Written by a skilled wordsmith, this marvelous opus invites viewers on an introspective trip, gently unraveling the veiled truths and profound impact resonating within the fabric of each and every word. Within the psychological depths of this touching review, we will embark upon a honest exploration of the book is primary subjects, dissect their fascinating writing fashion, and yield to the strong resonance it evokes heavy within the recesses of readers hearts.

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