Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, molar mass, mole, molecular mass, stoichiometry

	Get the Gizmo ready:	
Activity A:	 Check that the equation is still: 	
Moles	Fe ₂ O ₃ + 3CO 2Fe + 3CO ₂	144
	 If not, click New equation until it reappears. 	m.

Question: How do scientists find the molecular mass and molar mass of a substance?

 <u>Calculate</u>: The molecular mass of a molecule is the sum of the masses of each atom in the molecule. The unit of molecular mass is the universal mass unit (u).

Iron's atomic mass is 55.85 u, carbon's mass is 12.01 u, and oxygen's mass is 16.00 u.

- Calculate the molecular mass of carbon monoxide (CO) by adding the atomic mass of carbon and the atomic mass of oxygen:
- B. Calculate the molecular mass of carbon dioxide (CO₂): 44.01g
- Infer: A mole of a substance has a mass in grams that is equal to the molecular mass. For example, a carbon atom has a mass of 12.01 u. A mole of carbon has a mass of 12.01 g. Based on their molecular masses, list the molar mass of each substance:

Check your answers on the Gizmo by inspecting the middle row of tiles on the right side of the Gizmo. These tiles show the units "1 mol" on top and "g" below.

Practice: Hydrogen has an atomic mass of 1.01 u. What is the molar mass of these substances?

Exploration Sheet Answer Key Stoichiometry

Institute of Geological Sciences (Great Britain)

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Adopting the Melody of Term: An Emotional Symphony within Exploration Sheet Answer Key Stoichiometry

In some sort of eaten by screens and the ceaseless chatter of fast transmission, the melodic splendor and mental symphony produced by the prepared word often diminish into the backdrop, eclipsed by the relentless sound and interruptions that permeate our lives. But, situated within the pages of **Exploration Sheet Answer Key Stoichiometry** a marvelous fictional value filled with fresh emotions, lies an immersive symphony waiting to be embraced. Constructed by an elegant composer of language, this captivating masterpiece conducts viewers on an emotional trip, well unraveling the concealed melodies and profound affect resonating within each cautiously crafted phrase. Within the depths of this emotional examination, we shall discover the book is key harmonies, analyze their enthralling publishing model, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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