

$$M_{\text{AgNO}_3} = \frac{m_{\text{AgNO}_3}}{V_{\text{AgNO}_3}} = \frac{25.513 \text{ g} + 149.87 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{1.569 \frac{\text{mol}}{\text{L}}}$$

4. Confirm that the molarity is the same molarity that you calculated in step 3.

Yes, the molarity is the same as the molarity calculated in step 3.

### Part B: Unknown solutions

Using a method similar to the one used in part A, create and carry out a procedure for determining the molarity of the unknown  $\text{AgNO}_3$  solution and the unknown  $\text{NaCl}$  solution. Round your data and show your calculations.

#### Unknown $\text{AgNO}_3$

Upon the addition of solid  $\text{NaCl}$ , solid  $\text{AgCl}$  was formed.

$\text{AgCl}$  formed (g) = 2.32019 g

$$\text{mass}(\text{AgNO}_3) = \frac{M_r(\text{AgNO}_3)}{M_r(\text{AgCl})} \times \text{mass}(\text{AgCl}) = \frac{149.87}{143.32} \frac{\text{g}}{\text{mol}} \times 2.32019 \text{ g} = \boxed{2.3500 \text{ g}}$$

$$M_{\text{AgNO}_3} = \frac{m_{\text{AgNO}_3}}{V_{\text{AgNO}_3}} = \frac{2.3500 \text{ g} + 149.87 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{0.16189 \frac{\text{mol}}{\text{L}}}$$

#### Unknown $\text{NaCl}$

Upon the addition of solid  $\text{AgNO}_3$ , solid  $\text{AgCl}$  was formed.

$\text{AgCl}$  formed (g) = 10.7907 g

$$\text{mass}(\text{NaCl}) = \frac{M_r(\text{NaCl})}{M_r(\text{AgCl})} \times \text{mass}(\text{AgCl}) = \frac{58.443}{143.32} \frac{\text{g}}{\text{mol}} \times 10.7907 \text{ g} = \boxed{4.4002 \text{ g}}$$

$$M_{\text{NaCl}} = \frac{m_{\text{NaCl}}}{V_{\text{NaCl}}} = \frac{4.4002 \text{ g} + 58.443 \frac{\text{g}}{\text{mol}}}{100.00 \text{ mL}} + \frac{1000 \text{ mL}}{1 \text{ L}} = \boxed{0.75294 \frac{\text{mol}}{\text{L}}}$$

### Conclusion questions

1. If you have a company that is using this method to create  $\text{AgCl}$ , knowing that  $\text{AgNO}_3$  is an expensive chemical, what do you think would be a better limiting reagent to use,  $\text{AgNO}_3$  or  $\text{NaCl}$ ? Why?

# Experiment Limiting Reactant Answers

**JA Banks**



## Experiment Limiting Reactant Answers:

*Laboratory Manual for Principles of General Chemistry* Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

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CHEMISTRY HANDBOOK & STUDY GUIDE Gr11-12 NE Kevin Smith, 2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text

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reactions The editorial team have collected contributions from around the world and standardized them for publication Each experiment will explore a modern chemistry scenario such as sustainable chemistry application in the pharmaceutical industry catalysis and material sciences to name a few All the experiments will be complemented with a set of questions to challenge the students and a section for the instructors concerning the results obtained and advice on getting the best outcome from the experiment A section covering practical aspects with tips and advice for the instructors together with the results obtained in the laboratory by students has been compiled for each experiment Targeted at professors and lecturers in chemistry this useful text will provide up to date experiments putting the science into context for the students

**Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

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processes they are investigating rather than simply performing a laboratory experiment to the specifications set by the manual Each experiment begins with a problem scenario and ends with questions requiring feedback on the problem

The MCAT Chemistry Book Ajikumar Aryangat,2021-01-15 Comprehensive Rigorous Prep for MCAT Chemistry The MCAT Chemistry Book presents a comprehensive review of general chemistry and organic chemistry to prepare for the Medical College Admission Test Part I presents general chemistry concepts and Part II presents organic chemistry concepts The review sections are written in a user friendly manner to simplify and reduce the student s burden when deciphering difficult concepts At the end of each chapter practice questions are included to test the understanding of the key concepts Answers and explanations for the practice questions are provided after the review sections Illustrations and tables are included wherever necessary to focus and clarify key ideas and concepts

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