Experiment 25

Observations and Calculations: pH: Buffers and Their Properties

A. Determination of pH by the Use of Acid-Base Indicators

Color with 0.1 54 Solution of			
THE	NaH PO.	THUMO	ZaSO.
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Circle the observation(s) for each solution that was (were) most useful in estimating the pH range.

B. Measurement of the pH of Some Typical Solutions

Record the pH and the color observed with bromcresol green for each of the 0.1 M solutions that were tested.

pH A.9A 10.97 7.17 NaHSO. 10.97 T.17 LIFT

For any two solutions having a pH less than 6 or greater than 8, write a net ionic equation to explain qualitatively why the solution has that pH.

Solution NA 2 CO3 Equation $CO3^2 + H20 \Rightarrow HCO3 + OH^2$ Solution NA 11504 Equation $H509^- + H20 \Rightarrow (504)^2 + H30^+$

Explain why the color observed with bromcresol green for each of the four solutions is reasonable, given the pH.

C. Some Properties of Buffers

Buffer system selected HCO3 - CO3 HB is hydrogen cortonate - (or beneft (name the acid)

1. pH of buffer 9-29 [H*] 5.24 × 10 M pK, (by Eq. 4b)

2. pH of diluted buffer 9.24 [H*] 5.75 × 10 M pK,

pH after addition of 5 drops NaOH

pH after addition of 5 drops HCl

Comment on your observations in Parts 1 and 2.

Experiment 25ph Measurements Buffers And Their Properties

Allen Kennedy Herd

Experiment 25ph Measurements Buffers And Their Properties:

Invertebrate Oxygen-binding Proteins Jean Lamy, Josette Lamy, 1981 **Determination of the Size and Shape of Protein Molecules** P. Alexander, R. J. Block, 2014-05-12 Analytical Methods of Protein Chemistry Volume 3 Determination of the Size and Shape of Protein Molecules provides information pertinent to the analysis and isolation of protein This book deals with the measurement of the macromolecular properties of proteins Organized into seven chapters this volume begins with an overview of the theory and practice of the electron microscope to allow an understanding of the type of object that may be examined This text then describes the methods of making protein molecules conform to such an ideal which are the techniques of specimen preparation Other chapters consider the determinations of osmotic pressures of proteins This book discusses as well the experimental basis for the theory of the diffusion process in liquids The final chapter deals with the technical problem characteristics of light scattering This book is a valuable resource for electron microscopists protein chemists biologists physicist physico chemists scientists and research workers Russian Journal of Physical Chemistry Journal of Experimental Biology, 2005 The Journal of Biological Chemistry ,1965 Vols 3 140 include the .1963 society s Proceedings 1907 41 A Laboratory Manual of Analytical Methods of Protein Chemistry, Including Polypeptides Peter Alexander, Richard Joseph Block, Harold Palmer Lundgren, 1960 In the last fifteen years there has been a revolution in the techniques available for the analysis and isolation of proteins Every time a new technique has been introduced numerous papers have appeared describing modifications to it and the research worker who wishes to employ these methods is faced with a very serious problem in deciding which particular variant to use These volumes are intended to provide the fullest practical detail so that any scientist can follow the procedure by using this book alone and without having recourse to the original literature The techniques which are described in full are ones in which all the authors have had first hand experience and the descriptions contain those small but important points which save so much time In the first volume separation and isolation procedures are discussed the second concerns its analysis and reactivity and the third volume with the measurement of the macromolecular properties of proteins **Drug-like Properties: Concepts, Structure Design and Methods** Li Di, Edward H Kerns, 2010-07-26 Of the thousands of novel compounds that a drug discovery project team invents and that bind to the therapeutic target typically only a fraction of these have sufficient ADME Tox properties to become a drug product Understanding ADME Tox is critical for all drug researchers owing to its increasing importance in advancing high quality candidates to clinical studies and the processes of drug discovery If the properties are weak the candidate will have a high risk of failure or be less desirable as a drug product This book is a tool and resource for scientists engaged in or preparing for the selection and optimization process The authors describe how properties affect in vivo pharmacological activity and impact in vitro assays Individual drug like properties are discussed from a practical point of view such as solubility permeability and metabolic stability with regard to fundamental understanding applications of property data in

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