

- 2) Light the Bunsen burner and place it under the beaker. Adjust the burner so the hottest part of the flame is touching the bottom of the beaker.
- 3) Carefully heat and stir the mixture in the beaker using your stir rod. The solution should be hot, but not boiling. After all the crystals have dissolved, remove the beaker from the heat using your hot pad or rubber-coated tongs.
- 4) Add 2.00g of iron filings slowly to the hot copper (II) sulfate solution while stirring. Record observations in your data table.
- 5) Allow the beaker to cool for 10-15 minutes.
- 6) Pour off (decant) the solution into a different beaker. Pouring the solution down a stirring rod is recommended (as shown in the image shown to the right). Make sure not to disturb the copper in the beaker.
- 10) Add a small amount of water (10-15mL) to the copper & stir.
- 11) Let the copper settle and decant again.
- 12) Dry the copper as your teacher directs and determine its mass. Record the mass in the data table provided.



Data Table:

Mass of clean, dry 250mL beaker		2
Observations after adding iron filings		
	Circle the observations that support the chemical reaction.	
Mass of beaker with dried copper residue		2
Mass of copper residue		2

Analysis Questions:

- 1) Which reactant is limiting? What observations support this? Use the balanced chemical equation for the reaction that occurred in the lab and stoichiometric calculations to support your answer.
- 2) Why was the copper washed with water in step 10 of the procedure?
- 3) Why didn't the water added to the copper (II) sulfate need to be measured exactly?
- 4) What is the formula for the iron-containing compound that is formed when copper (II) sulfate and iron react? Support your answer.

Cleaning Up:

- 1) Place the copper in the waste container provided by your teacher.
- 2) Place the remaining solution in the labeled container.
- 3) Clean up all materials and return them to the proper locations.
- 4) Wash your hands thoroughly before leaving the lab.

Conclusions: Provide a summary of the lab and discuss a final confirmation or rejection of your hypothesis, citing your observations.

Experiment 37 Stoichiometry Answers

H Kauffman



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AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam

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perspective of ligand binding theory Review of selected examples that illustrate integration of structural and thermodynamic analysis Perfect for students and educators in chemistry biochemistry molecular biology and pharmaceutical science Ligand binding Basics will also appeal to practitioners who aim to study ligand binding in any molecular system **Chemistry, Study Guide** Bernice G. Segal, Peter S. Shenkin, 1989-02-14 This Second Edition of the first year chemistry text known for its clarity of exposition and its large number of illustrative worked problems contains a more rigorous treatment of electrochemistry chemical equilibrium and thermochemistry Worked examples now number over 300 and exercises over 1460 *Physiology of the Gastrointestinal Tract* Kim E. Barrett, Faye K. Ghishan, Juanita L. Merchant, Hamid M. Said, Jackie D. Wood, 2006-05-10 FROM THE PREFACE The original purpose of the First Edition of *Physiology of the Gastrointestinal Tract* to collect in one set of volumes the most current and comprehensive knowledge in our field was also the driving force for the Fourth Edition The explosion of information at the cellular level made possible in part by the continued emergence of powerful molecular and cellular techniques has resulted in a greater degree of revision than that of any other edition The first section now titled Basic Cell Physiology and Growth of the GI Tract contains numerous new chapters on topics such as transcriptional regulation signaling networks in development apoptosis and mechanisms in malignancies Most of the chapters in this section were edited by Juanita L Merchant Section II has been renamed Neural Gastroenterology and Motility and has been expanded from seven chapters with rather classic titles to more than twenty chapters encompassing not only the movement of the various parts of the digestive tract but also cell physiology neural regulation stress and the regulation of food intake Almost all of the chapters were recruited and edited by Jackie D Wood The third section is entirely new and contains chapters on Immunology and Inflammation which were edited by Kim E Barrett The fourth section on the Physiology of Secretion consists of chapters with familiar titles but with completely updated information to reflect the advances in our understanding of the cellular processes involved in secretion The last section on Digestion and Absorption contains new chapters on the intestinal barrier protein sorting and ion channels along with those focusing on the uptake of specific nutrients These chapters were recruited and edited by Hamid M Said and Faye K Ghishan Collected in one set the most current and comprehensive coverage of gastrointestinal physiology Information presented in a style that is both readable and understandable Valuable to the specialized researcher the clinical gastroenterologist the teacher and the student Features an entirely new section on Immunology and Inflammation Each section edited by the preeminent scientist in the field **100 questions and answers for job interview Offshore Drilling Platforms** PETROGAV INTERNATIONAL, This book offers you a brief but very involved look into the operations in the drilling of an oil gas wells that will help you to be prepared for job interview at oil gas companies From start to finish you ll see a general prognosis of the drilling process If you are new to the oil gas industry you ll enjoy having a leg up with the knowledge of these processes If you are a seasoned oil gas person you ll enjoy reading what you may or may not know in these pages This course provides a non technical

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Cancer as a Metabolic Disease Thomas Seyfried, 2012-05-18 The book addresses controversies related to the origins of cancer and provides solutions to cancer management and prevention It expands upon Otto Warburg's well known theory that all cancer is a disease of energy metabolism However Warburg did not link his theory to the hallmarks of cancer and thus his theory was discredited This book aims to provide evidence through case studies that cancer is primarily a metabolic disease requiring metabolic solutions for its management and prevention Support for this position is derived from critical assessment of current cancer theories Brain cancer case studies are presented as a proof of principle for metabolic solutions to disease management but similarities are drawn to other types of cancer including breast and colon due to the same cellular mutations that they demonstrate

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