

4. a. A sample of a $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$ solid salt mixture is dissolved in ~150 mL of deionized water previously adjusted to a pH that is basic. The precipitate, after having been filtered, was air-dried and weighed. Data for Trial 1 were obtained as shown. Complete the following table. (See *Report Sheet*.) Record calculated values with the correct number of significant figures.

A. Precipitation of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ from the Salt Mixture

Calculation Zone

3. Mass of salt mixture (g)	<u>0.879</u>
4. Mass of filter paper (g)	<u>1.896</u>
5. Mass of filter paper and $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u>2.180</u>
6. Mass of air-dried $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u> </u>

Data Analysis, 1.

B. Determination of Limiting Reactant

1. Limiting reactant in salt mixture	<u>$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$</u>
2. Excess reactant in salt mixture	<u> </u>

Data Analysis, 3.

Data Analysis

1. Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ precipitated (mol) <i>Show calculation.</i>	<u> </u>
2. Moles of limiting reactant in salt mixture (mol) <i>See equation 8.1.</i>	<u> </u>
3. Mass of limiting reactant in salt mixture (g) <i>Show calculation.</i>	<u> </u>
4. Mass of excess reactant in salt mixture (g) <i>Equals mass of salt mixture minus mass of limiting reactant. Show calculation.</i>	<u> </u>
5. Percent limiting reactant in salt mixture (%) <i>Show calculation.</i>	<u> </u>
6. Percent excess reactant in salt mixture (%)	<u> </u>

Data Analysis, 4.

Data Analysis, 5.

4. b. For Trials 2 and 3, the percent $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the mixture was 39.7% and 28.4% respectively.

a. What is the average percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? **Data Analysis, B.**

b. What are the standard deviation and the relative standard deviation (%RSD) for the percent of $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ in the sample? **Data Analysis, C and D.**

Exp 8 Limiting Reactant Prelab Answers

Wenbin Ji



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Table of Contents Exp 8 Limiting Reactant Prelab Answers

1. Understanding the eBook Exp 8 Limiting Reactant Prelab Answers
 - The Rise of Digital Reading Exp 8 Limiting Reactant Prelab Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Exp 8 Limiting Reactant Prelab Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Exp 8 Limiting Reactant Prelab Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Exp 8 Limiting Reactant Prelab Answers
 - Personalized Recommendations

- Exp 8 Limiting Reactant Prelab Answers User Reviews and Ratings
- Exp 8 Limiting Reactant Prelab Answers and Bestseller Lists
- 5. Accessing Exp 8 Limiting Reactant Prelab Answers Free and Paid eBooks
 - Exp 8 Limiting Reactant Prelab Answers Public Domain eBooks
 - Exp 8 Limiting Reactant Prelab Answers eBook Subscription Services
 - Exp 8 Limiting Reactant Prelab Answers Budget-Friendly Options
- 6. Navigating Exp 8 Limiting Reactant Prelab Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Exp 8 Limiting Reactant Prelab Answers Compatibility with Devices
 - Exp 8 Limiting Reactant Prelab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Exp 8 Limiting Reactant Prelab Answers
 - Highlighting and Note-Taking Exp 8 Limiting Reactant Prelab Answers
 - Interactive Elements Exp 8 Limiting Reactant Prelab Answers
- 8. Staying Engaged with Exp 8 Limiting Reactant Prelab Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Exp 8 Limiting Reactant Prelab Answers
- 9. Balancing eBooks and Physical Books Exp 8 Limiting Reactant Prelab Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Exp 8 Limiting Reactant Prelab Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Exp 8 Limiting Reactant Prelab Answers
 - Setting Reading Goals Exp 8 Limiting Reactant Prelab Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Exp 8 Limiting Reactant Prelab Answers
 - Fact-Checking eBook Content of Exp 8 Limiting Reactant Prelab Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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web for r12 saturation pressure at 120 f is 172 35 psia for r134a saturation pressure at 120 f is 186 023 psia if a lower operating pressure is the only desired criterion for the design clearly r12 is a better choice c arizona s summer the design specification given thus far is the temperature of the condenser see diagram which is 120 f