

First/Second Semester B.E. Degree Examination, June/July 2016
Engineering Chemistry

Time: 3 hrs.

Max. Marks: 80

Note: Answer any FIVE full questions, choosing one full question from each module.

Module-1

1.
 - a. Explain the construction and working of silver-silver chloride electrode. (05 Marks)
 - b. What are ion selective electrodes? Discuss the construction and working of a glass electrode. (06 Marks)
 - c. Explain the construction and working of Zinc-Air cell. (05 Marks)

OR

2.
 - a. Derive Nerst expression for single electrode potential. (05 Marks)
 - b. Explain the following battery characteristics :
 - i) Energy efficiency
 - ii) Shelf life
 - iii) Cycle life
 - c. Explain the construction and working of Methanol - Oxygen fuel cell with H_2SO_4 electrolyte. (06 Marks)

Module-2

3.
 - a. Explain electrochemical theory of corrosion with an example. (05 Marks)
 - b. Discuss the differential metal corrosion with an example. (05 Marks)
 - c. What is electroless plating? Mention the differences between electroplating and electroless plating. (06 Marks)

OR

4.
 - a. Discuss the following factors influencing nature of electrodeposit. (06 Marks)
 - i) Brighteners
 - ii) Levellers and
 - iii) pH
 - b. Explain electroplating of Decorative chromium. (05 Marks)
 - c. Discuss the process of Galvanization of Iron sheet. (05 Marks)

Module-3

5.
 - a. How Calorific value of a solid fuel is determined using bomb calorimeter? (05 Marks)
 - b. What is meant by cracking? Describe with a neat diagram, fluidized bed cracking method. (06 Marks)
 - c. Explain the construction and working of a photovoltaic cell. (05 Marks)

Engineering Chemistry 1st Year Chem Lab Manual

Daniel Meeroff



Engineering Chemistry 1st Year Chem Lab Manual:

The United States Catalog ,1903 **Engineering Chemistry Laboratory Manual** Daniel Meeroff,2013

Engineering Chemistry Laboratory Manual Dr Manoj Kumar Solanki,2019-03-20 Life is impossible without chemistry Engineering chemistry has a special role to play in the curriculum of under graduate students of all branches of Engineering The present book entitled ENGINEERING CHEMISTRY LABORATORY MANUAL is very useful to Engineering students of various Institutions The practical book providing simple and easy approach on the subject matter to Engineering students

Journal of the Engineering Institute of Canada ,1921 Vol 7 no 7 July 1924 contains papers prepared by Canadian engineers for the first World power conference July 1924 **Engineering Chemistry Laboratory Manual** Shirish Kumar KODADI,2020-08-31 Over the most recent couple of years the importance of undergraduate technical education has grown amid a huge industrial revolution in our country More refined and recently discovered super specific topics are being introduced instead of old ones while modifying the course curriculum In the new course curriculum more noteworthy accentuation is laid on the basic science subjects and on the need to develop in depth knowledge about the fundamentals of any particular area of academic interest Keeping all this in mind and utilizing my long experience as a teacher in a technical college under a technical university I have ventured to write this book titled Engineering Chemistry Laboratory Manual In this book all experiments are explained as per the JNTU syllabus for the first year students of B Tech These are supplemented with theoretical explanations followed by procedure description tabulation calculation sample calculation and finally a series of possible viva voce questions and their answers relating to that experiment This book will certainly help all B Tech B E students to do well in their viva voce while completing their experiments cum examinations It will also serve as a textbook in Chemistry practical examinations for any student in the laboratory I sincerely hope that this book will receive full appreciation from both students and teachers **The Engineering Journal** ,1921 **The Journal of the Engineering Institute of Canada** Engineering Institute of Canada,1921 **A Concise Engineering Chemistry Lab Manual for I/II Semester (I Year Mandatory Course) B.E Students** Dr. S. R. Pratap,Dr. S. Z. Mohamed Shamshuddin, **The United States catalog; books in print 1902** Marion E. Potter,1903 *The Cosmopolitan* ,1912

Reviewing **Engineering Chemistry 1st Year Chem Lab Manual**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Engineering Chemistry 1st Year Chem Lab Manual**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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