

PART A – CHEMISTRY

1. An unknown alcohol is treated with the "Lucas reagent" to determine whether the alcohol is primary, secondary or tertiary. Which alcohol reacts fastest and by what mechanism :
 - (1) tertiary alcohol by S_N1
 - (2) secondary alcohol by S_N2
 - (3) tertiary alcohol by S_N2
 - (4) secondary alcohol by S_N1
2. The first ionisation potential of Na is 5.1 eV. The value of electron gain enthalpy of Na^+ will be :
 - (1) - 5.1 eV
 - (2) - 10.2 eV
 - (3) + 2.55 eV
 - (4) - 2.55 eV
3. Stability of the species Li_2 , Li_2^- and Li_2^+ increases in the order of :
 - (1) $Li_2^- < Li_2^+ < Li_2$
 - (2) $Li_2 < Li_2^- < Li_2^+$
 - (3) $Li_2^- < Li_2 < Li_2^+$
 - (4) $Li_2 < Li_2^+ < Li_2^-$
4. The molarity of a solution obtained by mixing 750 mL of 0.5(M)HCl with 250 mL of 2(M)HCl will be :
 - (1) 1.00 M
 - (2) 1.75 M
 - (3) 0.975 M
 - (4) 0.875 M
5. Which of the following is the wrong statement ?
 - (1) O_3 molecule is bent.
 - (2) Ozone is violet-black in solid state.
 - (3) Ozone is diamagnetic gas.
 - (4) $ONCl$ and ONO^- are not isoelectronic.
6. Four successive members of the first row transition elements are listed below with atomic numbers. Which one of them is expected to have the highest $E^0_{M^{3+}/M^{2+}}$ value ?
 - (1) Mn(Z = 25)
 - (2) Fe(Z = 26)
 - (3) Co(Z = 27)
 - (4) Cr(Z = 24)
7. A solution of (-)-1-chloro-1-phenylethane in toluene racemises slowly in the presence of a small amount of $SbCl_5$ due to the formation of :
 - (1) carbene
 - (2) carbocation
 - (3) free radical
 - (4) carbanion
8. The coagulating power of electrolytes having ions Na^+ , Al^{3+} and Ba^{2+} for arsenic sulphide sol increases in the order :
 - (1) $Na^+ < Ba^{2+} < Al^{3+}$
 - (2) $Ba^{2+} < Na^+ < Al^{3+}$
 - (3) $Al^{3+} < Na^+ < Ba^{2+}$
 - (4) $Al^{3+} < Ba^{2+} < Na^+$

Engineering Chemistry Paper Solution 2013

Lingsheng Yao



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