



Nuclear Equations

(a) Disintegration of radium-236 by emission of an alpha particle (${}^4_2\text{He}$).



Mass Nos : Reactants = 236 ; Products = 232 + 4 = 236

Atomic Nos : Reactants = 88 ; Products = 86 + 2 = 88

(b) Disintegration of phosphorus-32 by emission of a beta particle (${}^0_{-1}\text{e}$).



Mass Nos : Reactants = 32 ; Products : 0 + 32 = 32

Atomic Nos : Reactants = 15 ; Products : 16 - 1 = 15

(c) Fission of argon-40 by bombardment with a proton (${}^1_1\text{H}$).



Mass Nos : Reactants = 40 + 1 = 41 ; Products = 40 + 1 = 41

Atomic Nos : Reactants = 18 + 1 = 19 ; Products = 19 + 0 = 19

Mcq Nuclear Equation

CL Gary



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