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Chy.

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CH₃ + iO-H
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3°-কার্বোনিয়ান আমন

দ্রু কর্মন্ত

CH₃C-O-H

চারসিয়ারি বিভিটাইল আলকোমল

 $S_{\rm N}1$ বিক্রিয়ার ক্লেক্সে RX এর সক্রিয়তার ক্রম হলো : 3 RX > 2 RX > 1 RX

 $S_N 2$ বা দ্বি-আণবিক নিউক্লিভফিলিক প্রতিছাপন : যে S_N বিক্রিয়ার হার বা গতি নিউক্লিভফাইল ও অ্যালকাইল হ্যালাইড (বা সাবস্টেট) উভয়ের ঘনমাত্রার ওপর নির্ভর করে, তাকে $S_N 2$ বিক্রিয়া বলে।

2

বৈশিষ্ট্য : (i) 1 হ্যালাইডে S_N2 বেশি ঘটে, কারণ এতে স্টেরিক বাধা কম; (ii) বিক্রিয়াটি এই নিউক্লিউল OH আয়নের ঘনমাত্রা বেশি থাকতে হয়; (iv) বিক্রিয়াটি ২য় ক্রমের হয় অর্থাৎ বিক্রিয়ার আয়ন উভয়ের ঘনমাত্রার ওপর নির্ভর করে। (v) S_N2 বিক্রিয়াটিতে সরাসরি নিউক্লিউলাইল OH আরল নতুন বন্ধন গঠন ও পুরাতন বন্ধন ভাঙ্গনজনিত অবছাত্তর অহায়ী জটিল সূহি হয়ন (vi) কার্বানিয়াম আয়ন সৃষ্টি হয় না; (vii) RX অণুর ক্রিয়াত্রিক গঠনে কার্বনের যে পার্শ্বে সর্বানিয়াম আয়ন সৃষ্টি হয় না; (vii) RX অণুর ক্রিয়াত্রিক গঠনে কার্বনের যে পার্শ্বে কাঠামো উল্টেখেক 180 অবছানে নিউক্লিওফাইল আক্রমণ করে। তাই উৎপাদ অণুর জ্ঞামিতিক কাঠামো উল্টেখ্য

হয়।

SN2 বিক্রিয়ার কৌশল: R-CH₂-X এর X পরমাণু যে দিকে থাকে, তার বিপরীত দিক ও বর্ষন পরমাণুকে নিউক্লিওফাইল (:OH) আক্রমণ করে। তথন নতুন সমযোজী বন্ধন গড়ন এবং পুর কর্ষন পরমাণুকে নিউক্লিওফাইল (:OH) আক্রমণ করে। তথন নতুন সমযোজী বন্ধন গড়ন এবং পুর সহকারে 'অহায়ী জটিল' সৃষ্টি হয়। শেষে C-X বন্ধনটি ভেঙ্গে গিয়ে হ্যালাইড (X) আয়ন মুক্ত হ পেকারে 'অহায়ী জটিল' সৃষ্টি হয়। শেষে ৫-X বন্ধনটি ভেঙ্গে গিয়ে হ্যালাইড (X) আয়ন মুক্ত হ পেরমাণুর সাথে পূর্ণ সমযোজী বন্ধনে আবন্ধ হয়ে পড়ে। তথন উৎপাদ অণুর কনফিগারেশনটি ৫ পরমাণুর সাথে পূর্ণ সমযোজী বন্ধনে আবন্ধ হয়ে পড়ে। তথন উৎপাদ অণুর কনফিগারেশনটি ৫ তিন পরমাণু বা মূলক ঝড়ে উল্টানো ছাতার মতো উল্টে গিয়ে নতুন ত্রিমাত্রিক গঠন লাভ করে। তিন পরমাণু বা মূলক ঝড়ে উল্টানো ছাতার মতো উল্টে গিয়ে লতুন ত্রিমাত্রিক গঠন লাভ করে। কাঠামো পরিবর্তনকে বিজ্ঞানী পল ওয়ালভেনের নামানুসারে ওয়ালভেন ইনভারসন (Walden inv

সক্রিয়তার ক্রম হলো 1° RX > 2° RX > 3°RX L এবং 2°RX এর বেলায় S_N1 ও S_N2 সমভাবে ঘটে

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Tobias Bleicker

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