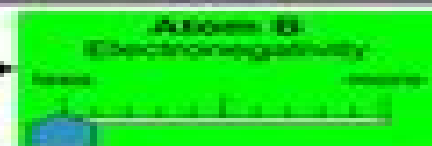


Change the electronegativity of atom B by dragging the slider all the way to less. →



Then, increase the electronegativity of atom c by dragging the slider so that it is half way between less and more →



4) Which atom has the highest electronegativity?

5) What happened to the molecular dipole arrow? Why does it point the way it points now?

6) Is the molecule polar? If not, why not? If so, which end is positive and which end is negative.

Increase the electronegativity of Atom A by dragging The slider so that it is half way between less and more →



7) What happened to the molecular dipole arrow?

8) Is the molecule polar and, if so, which end of the molecule has each charge. Explain.

High School Polarity Inquiry Activities Chemistry

Julia Schneider



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Problems and Problem Solving in Chemistry Education Georgios Tsapalis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner **Resources in Education** ,1996-10 *Current Index to Journals in Education* ,2002

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