

Objective: Students will identify and differentiate between the different parts and models of the atom.

1. Atoms are made of three basic particles: \_\_\_\_\_
2. What is electric charge? \_\_\_\_\_
3. A positive and a negative charge will \_\_\_\_\_
4. Two positive charges will \_\_\_\_\_
5. Two negative charges will \_\_\_\_\_
6. What does the term *isotope* represent? \_\_\_\_\_
7. Electric charge only appears in \_\_\_\_\_ of the elementary charges.
8. What did J.J. Thompson discover? \_\_\_\_\_
9. Atoms have \_\_\_\_\_ charge.
10. The model thought electrons were sprinkled around inside atoms like \_\_\_\_\_ in a \_\_\_\_\_ of \_\_\_\_\_.
11. This was the \_\_\_\_\_ ball model for the \_\_\_\_\_ of an \_\_\_\_\_, but it wasn't the \_\_\_\_\_ model.
12. In 1911 Rutherford, Geiger and Marsden did an experiment where they shot helium at a thin gold foil. They expected most of the helium to be deflected a little. What actually happened? \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
13. The best way to explain this was if a gold atom was mostly \_\_\_\_\_.
14. How much of an atom's mass is contained in the nucleus? \_\_\_\_\_
15. Where is the electron found? \_\_\_\_\_
16. Where is the proton found? \_\_\_\_\_
17. Where is the neutron found? \_\_\_\_\_
18. How much bigger is the electron cloud than the nucleus? \_\_\_\_\_
19. If the atom were a football stadium, how big would the nucleus be? \_\_\_\_\_  
How big would the electron be? \_\_\_\_\_
20. What is the charge of a proton? \_\_\_\_\_
21. What is the charge of an electron? \_\_\_\_\_
22. What is the charge of a neutron? \_\_\_\_\_
23. Why doesn't an electron fall into the nucleus? \_\_\_\_\_
24. What holds the nucleus together? \_\_\_\_\_
25. What are the four types of forces inside an atom? \_\_\_\_\_

# Foundations Of Physical Science Answers To Chapter 9

**Marco Cascella**



## **Foundations Of Physical Science Answers To Chapter 9:**

## **Foundations Of Physical Science Answers To Chapter 9** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Foundations Of Physical Science Answers To Chapter 9**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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two substances  $x_a$  and  $x_b$  the mole fractions of each would be written as

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12 4 355 154 g mol l

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web notice that the mole fraction has no units on it and is written as a decimal value do not change it to percent note of caution you could see the term mole percent it is simply the mole fraction multiplied by 100 for example in the problem just below the mole fraction of cinnamic acid is 0 2885 its mole percent would be 28 85

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