

<p style="text-align: center;">Rate and Growth</p> <p>Rate $\frac{dY}{dt}$</p> <p>Population Growth $\frac{dN}{dt} = B - D$</p> <p>Exponential Growth $\frac{dN}{dt} = r_{max} N$</p> <p>Logistic Growth $\frac{dN}{dt} = r_{max} N \left(\frac{K - N}{K} \right)$</p> <p>Temperature Coefficient Q_{10}[†] $Q_{10} = \left(\frac{k_2}{k_1} \right)^{\frac{10}{T_2 - T_1}}$</p> <p>Primary Productivity Calculation $\frac{\text{mg O}_2}{\text{L}} \times \frac{0.698 \text{ mL}}{\text{mg}} = \frac{\text{mL O}_2}{\text{L}}$ $\frac{\text{mL O}_2}{\text{L}} \times \frac{0.536 \text{ mg C fixed}}{\text{mL O}_2} = \frac{\text{mg C fixed}}{\text{L}}$ (at standard temperature and pressure)</p>	<p>dY = amount of change dt = change in time <i>B</i> = birth rate <i>D</i> = death rate <i>N</i> = population size <i>K</i> = carrying capacity <i>r_{max}</i> = maximum per capita growth rate of population</p> <p><i>T₂</i> = higher temperature <i>T₁</i> = lower temperature <i>k₂</i> = reaction rate at <i>T₂</i> <i>k₁</i> = reaction rate at <i>T₁</i> <i>Q₁₀</i> = the factor by which the reaction rate increases when the temperature is raised by ten degrees</p>	<p>Water Potential (Ψ) $\Psi = \Psi_p + \Psi_s$</p> <p>Ψ_p = pressure potential Ψ_s = solute potential</p> <p>The water potential will be equal to the solute potential of a solution in an open container because the pressure potential of the solution in an open container is zero.</p> <p>The Solute Potential of a Solution $\Psi_s = -iCRT$</p> <p><i>i</i> = ionization constant (this is 1.0 for sucrose because sucrose does not ionize in water) <i>C</i> = molar concentration <i>R</i> = pressure constant (<i>R</i> = 0.0831 liter bars/mole K) <i>T</i> = temperature in Kelvin ($^{\circ}\text{C} + 273$)</p>
<p style="text-align: center;">Surface Area and Volume</p> <p>Volume of a Sphere $V = \frac{4}{3} \pi r^3$</p> <p>Volume of a Rectangular Solid $V = lwh$</p> <p>Volume of a Right Cylinder $V = \pi r^2 h$</p> <p>Surface Area of a Sphere $A = 4\pi r^2$</p> <p>Surface Area of a Cube $A = 6s^2$</p> <p>Surface Area of a Rectangular Solid $A = \sum$ surface area of each side</p>	<p><i>r</i> = radius <i>l</i> = length <i>h</i> = height <i>w</i> = width <i>s</i> = length of one side of a cube <i>A</i> = surface area <i>V</i> = volume Σ = sum of all</p>	<p>Dilution (used to create a dilute solution from a concentrated stock solution) $C_i V_i = C_f V_f$</p> <p><i>i</i> = initial (starting) <i>C</i> = concentration of solute <i>f</i> = final (desired) <i>V</i> = volume of solution</p> <p>Gibbs Free Energy $\Delta G = \Delta H - T\Delta S$ ΔG = change in Gibbs free energy ΔS = change in entropy ΔH = change in enthalpy <i>T</i> = absolute temperature (in Kelvin)</p> <p>$\text{pH}^* = -\log_{10} [\text{H}^+]$</p>
<p>* For the purposes of the AP Exam, students will not be required to perform calculations using this equation; however, they must understand the underlying concepts and applications.</p> <p>† For use with labs only (optional).</p>		

Formula Biology

Tao Wei



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