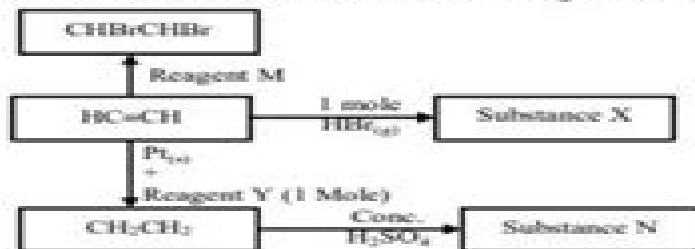


3. Ammonia is produced in large scale by Haber process.  
(i) Write an equation for the formation of ammonia gas.

(1 mark)

- (ii) State **two** optimum conditions for obtaining a high yield of ammonia in the process.  
(2 marks)

4. The scheme below shows some reactions starting with ethyne. Study it and answer the questions that follow.



- (a) Name substance X and N (1mark)

- (b) Name reagent M (1 Mark)

- (c) Ethene undergoes polymerization to form a polymer. Give an equation for the reaction and name the product.

- (i) Equation;

(1 mark)

- (ii) Name:

(1mark)

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