Question 4

4(a) Radiocarbon dating using ¹⁶C is a method of determining the age of once living, archaeological specimens which are no more than 50 000 years old.

123 is an example of a substance which can be used as a tracer. Tracers which concentrate in some organs are introduced into the bodies of living organisms and used to study chemical changes.

4(b)(i) Number of occupied electron shells = period number. Number of electrons in outer shell = group number.

	P	Q.	R	S
Group #	7	2	6	2
Period#	2	3	3	4

4(b)(ii) The compound is ionic.



Formula is QP2.

4(b)(iii) S will react more vigorously with hydrochloric acid than Q because S is lower in group 2 so it loses its outer electrons more easily than Q.

Equation for reaction of Q with HCI: $Q_{(s)} + 2HCI_{(aq)} \rightarrow QCI_{2(aq)} + H_{2(g)}$.

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Clemens Wendtner



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