



Student Exploration: Solubility and Temperature

Background: A material whose molecules, ions, or atoms are dissolved in a liquid is called a **solute**. The liquid is called the **solvent**.

Prior Knowledge: Speed of Sound (https://www.gln.org/curriculum/grade7/science/speed-of-sound/)

1. When sugar is added to water, it is eventually all sugar (dissolved water).
If you stir in a spoonful of sugar from the bottom, the sugar disappears before the water.

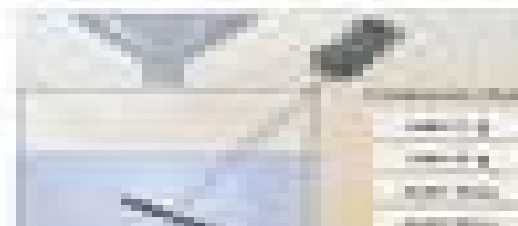
2. When sugar is added to water, it is **dissolved** (it enters the space between the glass and water).
A **homogeneous mixture** is formed. What is this mixture called?

3. If you stir in too much sugar, there will be undissolved sugar. _____ If you stir in a lot more sugar, _____
4. When sugar dissolves more easily at high temperatures than at low temperatures, it is called **soluble**.

Warm-Up

An **unsaturated** solution contains a lower amount of **solute** than is **possible** and is **stable** (the structure is **balanced**).
Now, the amount of sugar in a solution starts increasing from the surface water to the bottom water. The solution becomes **saturated** (it can't hold any more sugar).

Temperature affects the **solubility** of a substance. Solubility is the ability of a substance to dissolve in a solvent.



1. In this solution, identify the solute. _____ (The solvent is water.)

2. When the temperature of the solution increases, the solubility of the solute _____.
3. When the temperature of the solution decreases, the solubility of the solute _____.

Explore Learning Solubility And Temperature Answers

Walter A. Thurber, Mary C. Durkee



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