



Name:

Keith

Date:

Student Exploration: Boyle's Law and Charles's Law

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: absolute zero, Boyle's law, Charles's law, Gay-Lussac's law, Kelvin scale, pressure

Prior Knowledge Question (Do this BEFORE using the Gizmo.)

A small helium tank measures about two feet (60 cm) high. Yet it can fill over 50 balloons! How can such a small tank contain enough helium to fill so many balloons?

the volume of the gas decreases and the tank can contain all the helium gas

Gizmo Warm-up

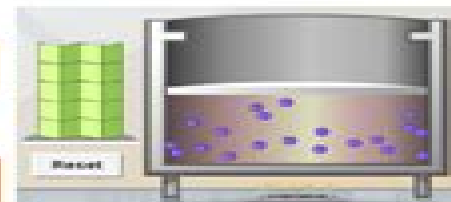
The Boyle's Law and Charles's Law Gizmo shows a container of gas. Inside, small purple spheres represent gas molecules.

1. Observe the particles. Are they all moving at the same speed?

No

2. How do the particles interact with the walls and lid of the container?

They bounce off of the lid and walls.



These interactions contribute to the **pressure** on the walls of the container. Pressure is defined as force per unit area. The SI units of pressure are newtons per square meter (N/m^2), or pascals (Pa).

2. Slowly drag the temperature (T) slider back and forth. (Note: In this Gizmo, the **Kelvin scale** is used to measure temperature. On the Kelvin scale, 0 degrees is **absolute zero**, the coldest possible temperature. Absolute zero is equal to -273.15°C or -459.67°F)

- A. How does the change in temperature affect the speed of the molecules?

It affects the pressure of the molecules.

- B. How does the change in temperature affect the volume of the container?

it affects how big or small the volume gets

Explore Learning Gizmo Answer Key Boyles Law

RJ Shavelson

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Explore Learning Gizmo Answer Key Boyles Law

1. Understanding the eBook Explore Learning Gizmo Answer Key Boyles Law
 - The Rise of Digital Reading Explore Learning Gizmo Answer Key Boyles Law
 - Advantages of eBooks Over Traditional Books
2. Identifying Explore Learning Gizmo Answer Key Boyles Law
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Gizmo Answer Key Boyles Law
 - User-Friendly Interface
4. Exploring eBook Recommendations from Explore Learning Gizmo Answer Key Boyles Law

- Personalized Recommendations
 - Explore Learning Gizmo Answer Key Boyles Law User Reviews and Ratings
 - Explore Learning Gizmo Answer Key Boyles Law and Bestseller Lists
5. Accessing Explore Learning Gizmo Answer Key Boyles Law Free and Paid eBooks
 - Explore Learning Gizmo Answer Key Boyles Law Public Domain eBooks
 - Explore Learning Gizmo Answer Key Boyles Law eBook Subscription Services
 - Explore Learning Gizmo Answer Key Boyles Law Budget-Friendly Options
 6. Navigating Explore Learning Gizmo Answer Key Boyles Law eBook Formats
 - ePub, PDF, MOBI, and More
 - Explore Learning Gizmo Answer Key Boyles Law Compatibility with Devices
 - Explore Learning Gizmo Answer Key Boyles Law Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Gizmo Answer Key Boyles Law
 - Highlighting and Note-Taking Explore Learning Gizmo Answer Key Boyles Law
 - Interactive Elements Explore Learning Gizmo Answer Key Boyles Law
 8. Staying Engaged with Explore Learning Gizmo Answer Key Boyles Law
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Gizmo Answer Key Boyles Law
 9. Balancing eBooks and Physical Books Explore Learning Gizmo Answer Key Boyles Law
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Gizmo Answer Key Boyles Law
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Explore Learning Gizmo Answer Key Boyles Law
 - Setting Reading Goals Explore Learning Gizmo Answer Key Boyles Law
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Explore Learning Gizmo Answer Key Boyles Law

- Fact-Checking eBook Content of Explore Learning Gizmo Answer Key Boyles Law
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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