Name: Date:

Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

If Nate were to give Clara one of his markers, then both would end up with eight.

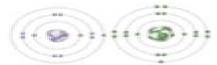
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

If Maggie gives Fred and Florence one marker each, they will all have eight markers.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lanic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (CI) are selected from the menus at right. Click Play (- to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



 Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons are called valence electrons.

How many valence electrons does each atom have? Sodium: 1 Chlorine:

7

Click Pause ("). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.

A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

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C Cleary

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Table of Contents Explore Learning Covalent Bonds Gizmo Answers

- 1. Understanding the eBook Explore Learning Covalent Bonds Gizmo Answers
 - The Rise of Digital Reading Explore Learning Covalent Bonds Gizmo Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Explore Learning Covalent Bonds Gizmo Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Covalent Bonds Gizmo Answers
 - User-Friendly Interface

- 4. Exploring eBook Recommendations from Explore Learning Covalent Bonds Gizmo Answers
 - Personalized Recommendations
 - Explore Learning Covalent Bonds Gizmo Answers User Reviews and Ratings
 - Explore Learning Covalent Bonds Gizmo Answers and Bestseller Lists
- 5. Accessing Explore Learning Covalent Bonds Gizmo Answers Free and Paid eBooks
 - o Explore Learning Covalent Bonds Gizmo Answers Public Domain eBooks
 - Explore Learning Covalent Bonds Gizmo Answers eBook Subscription Services
 - Explore Learning Covalent Bonds Gizmo Answers Budget-Friendly Options
- 6. Navigating Explore Learning Covalent Bonds Gizmo Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Explore Learning Covalent Bonds Gizmo Answers Compatibility with Devices
 - Explore Learning Covalent Bonds Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Covalent Bonds Gizmo Answers
 - Highlighting and Note-Taking Explore Learning Covalent Bonds Gizmo Answers
 - Interactive Elements Explore Learning Covalent Bonds Gizmo Answers
- 8. Staying Engaged with Explore Learning Covalent Bonds Gizmo Answers
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Covalent Bonds Gizmo Answers
- 9. Balancing eBooks and Physical Books Explore Learning Covalent Bonds Gizmo Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Covalent Bonds Gizmo Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Covalent Bonds Gizmo Answers
 - Setting Reading Goals Explore Learning Covalent Bonds Gizmo Answers
 - Carving Out Dedicated Reading Time

- 12. Sourcing Reliable Information of Explore Learning Covalent Bonds Gizmo Answers
 - Fact-Checking eBook Content of Explore Learning Covalent Bonds Gizmo Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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