Name

Sharanya Saxena

Date:

16 October, 2020

Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Since there are 8 markers in a set, Nate would have to give one marker to Clara:

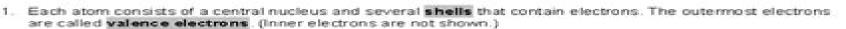
Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give up two markers, one of which can go to each Fred and Florence.

Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lonic Bonds* Gizmo allows you to explore how **lonic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (•) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

- - A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Sodium Which is a nonmetal? Chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

Sodium lets go of its one electron.

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Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents Computational Approaches for Chemistry Under Extreme Conditions Nir Goldman, 2019-02-18 This book presents recently developed computational approaches for the study of reactive materials under extreme physical and thermodynamic conditions It delves into cutting edge developments in simulation methods for reactive materials including quantum calculations spanning nanometer length scales and picosecond timescales to reactive force fields coarse grained approaches and machine learning methods spanning microns and nanoseconds and beyond These methods are discussed in the context of a broad range of fields including prebiotic chemistry in impacting comets studies of planetary interiors high pressure synthesis of new compounds and detonations of energetic materials. The book presents a pedagogical approach for these state of the art approaches compiled into a single source for the first time Ultimately the volume aims to make valuable research tools accessible to experimentalists and theoreticians alike for any number of scientific efforts spanning many different types of compounds and reactive conditions Lehninger Principles of Biochemistry David L. Nelson, Albert L. Lehninger, Michael M. Cox, 2008-02 Authors Dave Nelson and Mike Cox combine the best of the laboratory and best of the classroom introducing exciting new developments while communicating basic principles of biochemistry

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