



A PERIODIC TABLE LOGIC PROBLEM

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INTRODUCTION

- Analyze properties of elements using the periodic table.
- Identify elements based on their position in the periodic table.

INFORMATION PROVIDED

Information is provided for the 18 elements in the first two rows of the periodic table. These elements are listed in groups. The first group contains hydrogen, which is a nonmetal and most reactive element.

The first column of the periodic table contains the second-period elements. Hydrogen is the first element in this column.

Information is provided for the second-period elements. These elements are listed in groups. Hydrogen is the first element in the second-period group.

Information is provided for the third-period elements. Hydrogen is the first element in the third-period group. The second-period group contains the second-period elements. Hydrogen is the first element in the second-period group. The third-period group contains the third-period elements. Hydrogen is the first element in the third-period group. The fourth-period group contains the fourth-period elements. Hydrogen is the first element in the fourth-period group. The fifth-period group contains the fifth-period elements. Hydrogen is the first element in the fifth-period group. The sixth-period group contains the sixth-period elements. Hydrogen is the first element in the sixth-period group.

Information is provided for the seventh-period elements. Hydrogen is the first element in the seventh-period group. The eighth-period group contains the eighth-period elements. Hydrogen is the first element in the eighth-period group.

Information is provided for the ninth-period elements. Hydrogen is the first element in the ninth-period group. The tenth-period group contains the tenth-period elements. Hydrogen is the first element in the tenth-period group.

Information is provided for the eleventh-period elements. Hydrogen is the first element in the eleventh-period group.

Experiment 9 A Periodic Table Logic Problem

Mark Steyvers

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