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# Excel 20training Unit 1

**Thomas Griffiths**

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 $3\text{Ca(OH)}_2 + 1\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$  type of reaction double displacement  
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 $2\text{Al(OH)}_3 + 3\text{CaSO}_4 \rightarrow 2\text{Al}_2\text{SO}_4 + 3\text{Ca(OH)}_2$  type of reaction double displacement  
 $3\text{Mg} + 1\text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$  type of reaction single displacement

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$  type of reaction double displacement  
 $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$  type of reaction combustion  
 $2\text{Fe} + 6\text{NaBr} \rightarrow 2\text{FeBr}_3 + 6\text{Na}$  type of reaction single displacement  
 $\text{MgSO}_4 + 2\text{NH}_4\text{OH} \rightarrow \text{Mg(OH)}_2 + (\text{NH}_4)_2\text{SO}_4$  type of reaction double displacement  
 $\text{C}_4\text{H}_8 + 6\text{O}_2 \rightarrow 4\text{CO}_2 + 4\text{H}_2\text{O}$  type of reaction combustion  
 $\text{HCl} + \text{NaOH} \rightarrow \text{H}_2\text{O} + \text{NaCl}$  type of reaction double displacement

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