

# DE LA SALLE HOLY CROSS COLLEGE



## November Examinations 2014 Life Science Paper 1 Grade 11

DATE: 17 November 2014

MARKS: 200

EXAMINER: Mrs. Preston-Whyte

TIME: 3 HOURS

MODERATOR: A. Lawless

**INSTRUCTIONS:**

1. Read the questions carefully.
2. This paper consists of 22 pages including an Answer Booklet. Please check that your question paper is complete.
3. This question paper consists of 4 questions.
4. Question 1 must be answered in the Answer Booklet.
5. Questions 2, 3 and 4 must be answered on the folio paper.
6. Answer all questions.
7. Number your answers to the questions exactly as they are numbered in the question paper.
8. Do not use TIPPEX.
9. Write the question number in the middle of the page and rule off after each question.
10. Draw a marking margin on the right of the page.

**LEARNER NAME:** \_\_\_\_\_

| Aim                | Question No. | Marks Allocated |
|--------------------|--------------|-----------------|
| Aim 1: Theory      | 1.1-1.6      | 32              |
|                    | 2.1          | 7               |
|                    | 3.1-3.2      | 25              |
|                    | 2.2-2.3      | 31              |
|                    | 4.1-4.3      | 19              |
|                    | 4.4          | 6               |
| Aim 2: Practical   | 1.2-1.6      | 24              |
|                    | 2.1          | 8               |
|                    | 2.2-2.3      | 4               |
|                    | 3.1          | 7               |
| Aim 3: Application | 1.2          | 32              |
|                    | 3.2          | 8               |

# Exampler Life Science Paper 2 2014 Grade 11

**Hongru Du**



**Exampler Life Science Paper 2 2014 Grade 11:**

This book delves into Exemplar Life Science Paper 2 2014 Grade 11. Exemplar Life Science Paper 2 2014 Grade 11 is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Exemplar Life Science Paper 2 2014 Grade 11, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Exemplar Life Science Paper 2 2014 Grade 11
    - Chapter 2: Essential Elements of Exemplar Life Science Paper 2 2014 Grade 11
    - Chapter 3: Exemplar Life Science Paper 2 2014 Grade 11 in Everyday Life
    - Chapter 4: Exemplar Life Science Paper 2 2014 Grade 11 in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, this book will provide an overview of Exemplar Life Science Paper 2 2014 Grade 11. This chapter will explore what Exemplar Life Science Paper 2 2014 Grade 11 is, why Exemplar Life Science Paper 2 2014 Grade 11 is vital, and how to effectively learn about Exemplar Life Science Paper 2 2014 Grade 11.
  3. In chapter 2, this book will delve into the foundational concepts of Exemplar Life Science Paper 2 2014 Grade 11. The second chapter will elucidate the essential principles that must be understood to grasp Exemplar Life Science Paper 2 2014 Grade 11 in its entirety.
  4. In chapter 3, the author will examine the practical applications of Exemplar Life Science Paper 2 2014 Grade 11 in daily life. The third chapter will showcase real-world examples of how Exemplar Life Science Paper 2 2014 Grade 11 can be effectively utilized in everyday scenarios.
  5. In chapter 4, this book will scrutinize the relevance of Exemplar Life Science Paper 2 2014 Grade 11 in specific contexts. This chapter will explore how Exemplar Life Science Paper 2 2014 Grade 11 is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, the author will draw a conclusion about Exemplar Life Science Paper 2 2014 Grade 11. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Exemplar Life Science Paper 2 2014 Grade 11.

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$   
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$   
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$   
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$   
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web thus for reaction 9 2 in which 1 mole of  $\text{H}_3\text{O}^+$  and 1 mole of  $\text{OH}^-$  combine to form 2 moles of  $\text{H}_2\text{O}$   $\Delta h = 55.8 \text{ kJ}$  as shown note that the enthalpy change  $\Delta h_{\text{products}} - \Delta h_{\text{reactants}}$  is positive if heat is absorbed that is if  $\Delta h_{\text{products}}$

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