

Section C/B74

CHEMISTRY
Paper 6: Synthesis

[30]

Total: 30

Max. Marks: 100

Time: 1.5 Hrs.

General Instructions:

- (i) All questions are compulsory.
- (ii) Questions number 1 to 5 are very short answer questions and carry 1 mark each.
- (iii) Questions number 6 to 10 are short answer questions and carry 2 marks each.
- (iv) Questions number 11 to 12 are short answer questions and carry 3 marks each.
- (v) Questions number 13 to 14 are short answer questions and carry 4 marks each.
- (vi) Questions number 15 to 16 are long answer questions and carry 5 marks each.
- (vii) Use log table if necessary. Use of calculator is not allowed.

1. Write the formulae of the compound of phosphorus which is obtained by reacting PCl_5 with H_2O . (2)

Ans. H_3PO_4 , phosphoric acid

or, $\text{P} = \text{O}$, $\text{O} = \text{P}(\text{OH})_2$, $\text{O} = \text{P}(\text{OH})_2$, $\text{O} = \text{P}(\text{OH})_2$

2. Write the hybridisation of the following compound. (2)



Ans. $\text{H}_2\text{C}=\text{C} \begin{array}{c} \text{H} \\ | \\ \text{H} \end{array} \text{C} \begin{array}{c} \text{H} \\ | \\ \text{H} \end{array} \text{H}$



3. What is the effect of adding a catalyst? (2)

- (i) Activation energy (E_a) will
- (ii) Decrease (E_a) will increase.

Ans. (i) Catalyst generally is not required because its addition is found to be unnecessary. Hence catalyst is introduced only if required to decrease the activation energy.

(ii) Catalyst will decrease the activation energy of the reaction and increase the equilibrium constant of the reaction. It will not affect the equilibrium constant of the reaction.

4. One of the  and  is an example of a functional group. (2)

Ans.  ethylbenzene

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