

ANSWER KEY

ATOMS AND ELECTRONS STUDY GUIDE

Learning Goals: (1) (10-11 P. 41) Explain a new scientific theory of atomic-scale phenomena (atomic theory) by describing changes in the atomic model over time, including those changes that are consistent with experimental evidence. (10-11 P. 42) Explain the scientific theory of atomic-scale phenomena, including the scientific theories in terms of particles, motions, interactions, and the evidence supporting those particles and motions. (10-11 P. 42) Explain changes and motions within the atom. (10-11 P. 42) Explain properties of atoms and their motions within particles, including the composition of their components. (10-11 P. 42) Explain the scientific evidence of the first fundamental principles (principles) of atomic structure, such as: (1) atomic theory, (2) atomic theory, (3) atomic theory.

ANSWER KEY

Directions: Read the atomic theory selected on the correct reading.

ANSWER	ANSWER KEY	ANSWER KEY: ANSWER KEY
1C	Correctness (10-11 P. 41)	1. Everything is made of atoms (atoms) (10-11 P. 41) (10-11 P. 41)
2A	Correct (10-11 P. 41)	2. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)
3B	Correct (10-11 P. 41)	3. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)
4C	Correct (10-11 P. 41)	4. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)
5A	Correct (10-11 P. 41)	5. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)
6B	Correct (10-11 P. 41)	6. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)
7C	Correct (10-11 P. 41)	7. The atomic theory (10-11 P. 41) (10-11 P. 41) (10-11 P. 41)

Directions: Use the provided evidence to answer the following questions:

- 10-11 P. 41** The number of protons equals the number of electrons.
- 10-11 P. 41** The number of protons equals the number of electrons.
- 10-11 P. 41** The number of protons equals the number of electrons.

Energy And Electrons Study Guide Answers

Michael Hoelscher



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