

Ionic Compounds: Names and Formulas Worksheet

1. Write the formulas for the following compounds.

- (a) sodium fluoride NaF
 (b) potassium chloride KCl
 (c) calcium oxide CaO
 (d) lithium bromide LiBr
 (e) beryllium chloride BeCl_2
 (f) potassium sulfide K_2S
 (g) magnesium nitride Mg_3N_2

- (h) calcium phosphide Ca_3P_2
 (i) magnesium oxide MgO
 (j) aluminum chloride AlCl_3
 (k) sodium iodide NaI
 (l) potassium nitride K_3N
 (m) lithium sulfide Li_2S
 (n) barium phosphide Ba_3P_2

2. Write the names for the following compounds.

- (a) LiCl lithium chloride
 (b) MgS Magnesium sulfide
 (c) NaCl sodium chloride
 (d) Al_2O_3 aluminum oxide
 (e) CaS calcium sulfide
 (f) Zn_3P_2 zinc phosphide
 (g) Be_3P_2 beryllium phosphide

- (h) Na_2O sodium oxide
 (i) KBr potassium bromide
 (j) BeO beryllium oxide
 (k) CaF_2 calcium fluoride
 (l) BaBr_2 barium bromide
 (m) Na_3N sodium nitride
 (n) Al_2S_3 aluminum sulfide

Naming Worksheet 5 Ionic Compounds Summary

Answer Key

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Lodge, David Sang**



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