

## **Naming Ionic Compounds – Answer Key**

*Give the name of the following ionic compounds:*

		<b>Name</b>
1)	$\text{Na}_2\text{CO}_3$	<b>sodium carbonate</b>
2)	$\text{NaOH}$	<b>sodium hydroxide</b>
3)	$\text{MgBr}_2$	<b>magnesium bromide</b>
4)	$\text{KCl}$	<b>potassium chloride</b>
5)	$\text{FeCl}_2$	<b>iron (II) chloride</b>
6)	$\text{FeCl}_3$	<b>iron (III) chloride</b>
7)	$\text{Zn(OH)}_2$	<b>zinc hydroxide</b>
8)	$\text{Be}_2\text{SO}_4$	<b>beryllium sulfate</b>
9)	$\text{CrF}_2$	<b>chromium (II) fluoride</b>
10)	$\text{Al}_2\text{S}_3$	<b>aluminum sulfide</b>
11)	$\text{PbO}$	<b>lead (II) oxide</b>
12)	$\text{Li}_3\text{PO}_4$	<b>lithium phosphate</b>
13)	$\text{TiI}_4$	<b>titanium (IV) iodide</b>
14)	$\text{Co}_3\text{N}_2$	<b>cobalt (II) nitride</b>
15)	$\text{Mg}_3\text{P}_2$	<b>magnesium phosphide</b>
16)	$\text{Ga(NO}_2)_3$	<b>gallium nitrite</b>
17)	$\text{Ag}_2\text{SO}_3$	<b>silver sulfite</b>
18)	$\text{NH}_4\text{OH}$	<b>ammonium hydroxide</b>
19)	$\text{Al(CN)}_3$	<b>aluminum cyanide</b>
20)	$\text{Be(CH}_3\text{COO)}_2$	<b>beryllium acetate</b>

# Naming Ionic Compounds Key

**Pamela Mills, Amina El-Ashmawy**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author names.

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### Table of Contents **Naming Ionic Compounds Key**

1. Understanding the eBook **Naming Ionic Compounds Key**
  - The Rise of Digital Reading **Naming Ionic Compounds Key**
  - Advantages of eBooks Over Traditional Books
2. Identifying **Naming Ionic Compounds Key**
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an **Naming Ionic Compounds Key**
  - User-Friendly Interface
4. Exploring eBook Recommendations from **Naming Ionic Compounds Key**
  - Personalized Recommendations
  - **Naming Ionic Compounds Key** User Reviews and Ratings
  - **Naming Ionic Compounds Key** and Bestseller Lists

5. Accessing Naming Ionic Compounds Key Free and Paid eBooks
  - Naming Ionic Compounds Key Public Domain eBooks
  - Naming Ionic Compounds Key eBook Subscription Services
  - Naming Ionic Compounds Key Budget-Friendly Options
6. Navigating Naming Ionic Compounds Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Naming Ionic Compounds Key Compatibility with Devices
  - Naming Ionic Compounds Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Naming Ionic Compounds Key
  - Highlighting and Note-Taking Naming Ionic Compounds Key
  - Interactive Elements Naming Ionic Compounds Key
8. Staying Engaged with Naming Ionic Compounds Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Naming Ionic Compounds Key
9. Balancing eBooks and Physical Books Naming Ionic Compounds Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Naming Ionic Compounds Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Naming Ionic Compounds Key
  - Setting Reading Goals Naming Ionic Compounds Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Naming Ionic Compounds Key
  - Fact-Checking eBook Content of Naming Ionic Compounds Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning



- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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