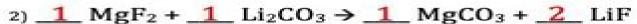
# **Balancing Chemical Reactions - Practice**

Balance the following reactions and identify what type of reactions they are.

1) 
$$1 N_2 + 3 H_2 \rightarrow 2 NH_3$$

Type of Reaction? \_\_\_\_ Synthesis



Type of Reaction? Double Replacement

3) 
$$1 P_4 + 3 O_2 \rightarrow 2 P_2O_3$$

Type of Reaction? Synthesis

4) 
$$2 \text{ RbNO}_3 + 1 \text{ BeF}_2 \rightarrow 1 \text{ Be(NO}_3)_2 + 2 \text{ RbF}_3$$

Type of Reaction? \_\_\_\_ Double Replacement

5) 
$$2 \text{ AgNO}_3 + 1 \text{ Cu} \rightarrow 1 \text{ Cu(NO}_3)_2 + 2 \text{ Ag}$$

Type of Reaction? Single Replacement

6) 
$$2 H_2O_2 \rightarrow 2 H_2O + 1 O_2$$

Type of Reaction? Decomposition

7) 2 HCN + 1 CuSO<sub>4</sub> 
$$\rightarrow$$
 1 H<sub>2</sub>SO<sub>4</sub> + 1 Cu(CN)<sub>2</sub>

Type of Reaction? \_\_\_\_ Double Replacement

8) 
$$\underline{1}$$
 GaF<sub>3</sub> +  $\underline{3}$  Cs  $\rightarrow$   $\underline{3}$  CsF +  $\underline{1}$  Ga

Type of Reaction? Single Replacement

# **Naming Chemical Reactions Practice**

John A. Olmsted, Gregory M. Williams, Robert Charles Burk

#### **Naming Chemical Reactions Practice:**

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