	レビン	
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# Assignment #2 – Compound Names and Formulas Multi-valent ions only

A. Write the correct formula for the	e following compounds, all of which have been named using the
	The ionic charge is given after the first element (metallic) in Roman
Numerals.	

1. copper (II) oxide CUO	11. manganese (III) oxide Mv 203		
2. mercury (I) oxide HQ20	12. vanadium (II) bromide VB(z		
3. gold (III) chloride _AuCl3	13. niobium(V) oxide Nb205		
4. thallium (III) bromide TIPY 3	14. titanium (III) oxide		
5. bismuth (V) oxide	15. titanium (III) nitride TIN		
6. terbium (III) oxide	16. iron (II) oxide <u>Fe.O</u>		
7. uranium (VI) oxide \( \lambda \ \O \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	17. cobalt (II) phosphide C03P2		
8. protactinium (IV) bromide PhBY4	18. tin (II) oxide		
9. cerium (III) oxide	19. thulium (II) bromide TWBY 2		
10. arsenic (V) sulphideAS_2S_5	20. copper (II) bromide CUPY2		
B. Name the compound using the Roman Numeral Method and balance the ionic charges.			
1. SnCl Tin (IV) Chloride	9. Pdf. Palladium (IV) Fluoride		
2. BIBIS BISMINTO STOMICE	10. OszOs OSMIUM (III) Oxide		
3. POO2 ROLONIUM (IV) OXICE	11. MOBEZ MOLYBORENIUM (II) Bramide		
4. Pbl2 100d(II) Iodide	12. VCIs Vanadium (V) Chloride		
5. HgO <u>Mercury</u> (II) Oxide	13. Mn2O3 Manganese (III) Oxide		
6. HgCI MEYCUYY (I) Chroride	14. COO COLONY (II) Oxide		
7. Auzos Gold (III) Oxide	15. Np2O3 NEDTUNIUM (II) OXICE		
8. FeClo Iron (II) Chloride	16. V2O6 Vanadium (V) Oxide		

# Naming Balancing And Formula Writing Answer Key

**Marian DeWane** 

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